



Search Report

EIC 1700

STIC Database Tracking Number: 245762

To: RIP LEE
Location: REM-10A21
Art Unit: 1796
Thursday, December 13, 2007

Case Serial Number: 10/524216

From: MEI HUANG
Location: EIC1700
REM-4B28 / REM-4B31
Phone: (571)272-3952

mei.huang@uspto.gov

Search Notes

Examiner LEE:

Please feel free to contact me if you have any questions or if you would like to refine the search query. Thank you for using STIC services!

Regards,
Mei



=> fil reg
FILE 'REGISTRY' ENTERED AT 16:22:43 ON 13 DEC 2007
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STRUCTURE FILE UPDATES: 12 DEC 2007 HIGHEST RN 957825-32-0
DICTIONARY FILE UPDATES: 12 DEC 2007 HIGHEST RN 957825-32-0

New CAS Information Use Policies, enter HELP USAGETERMS for details.

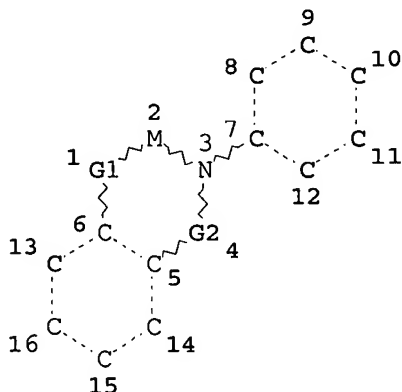
TSCA INFORMATION NOW CURRENT THROUGH June 29, 2007

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REGISTRY includes numerically searchable data for experimental and
predicted properties as well as tags indicating availability of
experimental property data in the original document. For information
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<http://www.cas.org/support/stngen/stndoc/properties.html>

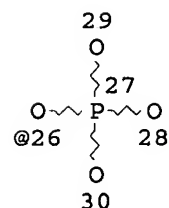
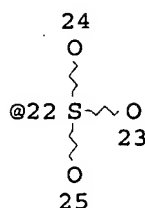
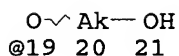
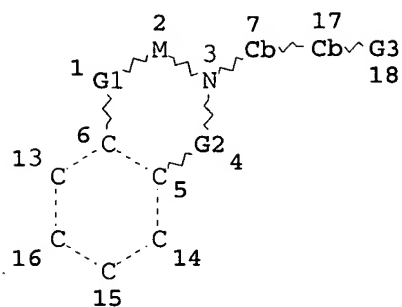
=> d que stat 125
L3 STR



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DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 16

STEREO ATTRIBUTES: NONE
L7 10084 SEA FILE=REGISTRY SSS FUL L3
L19 STR



@31 N +

VAR G1=O/S/N

VAR G2=C/N

VAR G3=19/22/26/31

NODE ATTRIBUTES:

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GGCAT IS MCY UNS AT 17

DEFAULT ECLEVEL IS LIMITED

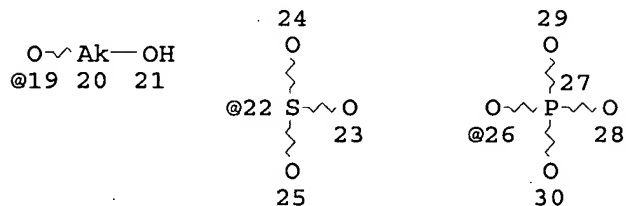
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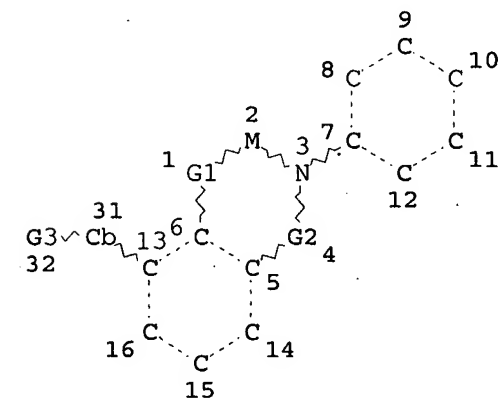
NUMBER OF NODES IS 26

STEREO ATTRIBUTES: NONE

L20 STR



@33 N +



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VAR G2=C/N

VAR G3=19/22/26/33

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CHARGE IS *+ AT 33

DEFAULT MLEVEL IS ATOM

GGCAT IS MCY UNS AT 31

DEFAULT ECLEVEL IS LIMITED

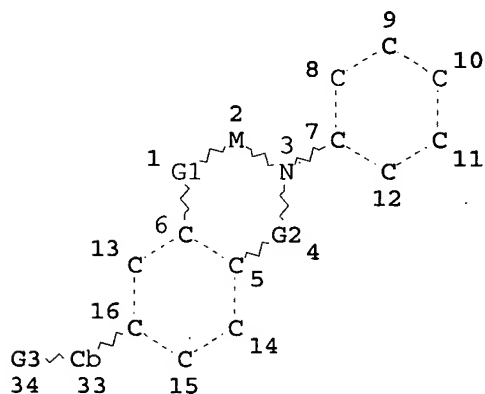
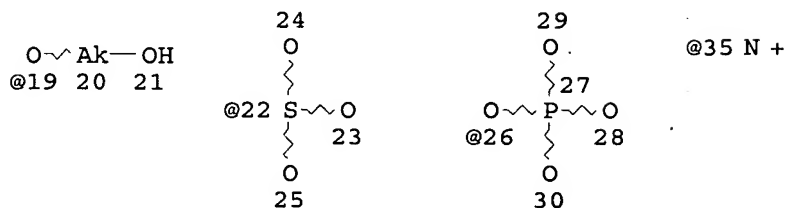
GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 31

STEREO ATTRIBUTES: NONE

L21 STR



VAR G1=O/S/N

VAR G2=C/N

VAR G3=19/22/26/35

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CHARGE IS *+ AT 35

DEFAULT MLEVEL IS ATOM

GGCAT IS MCY UNS AT 33

DEFAULT ECLEVEL IS LIMITED

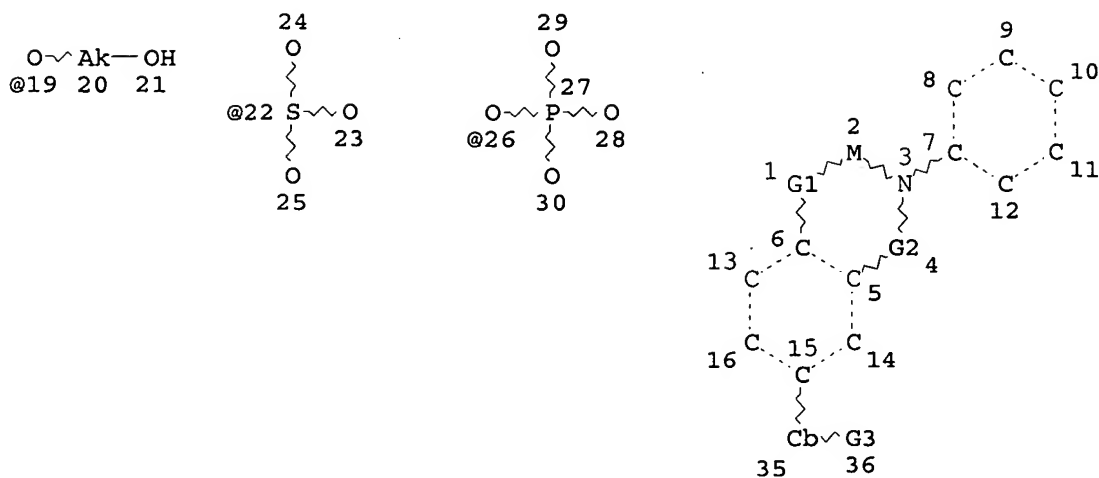
GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 31

STEREO ATTRIBUTES: NONE

L22 STR



@37 N +

VAR G1=O/S/N

VAR G2=C/N

VAR G3=19/22/26/37

NODE ATTRIBUTES:

CHARGE IS *+ AT 37

DEFAULT MLEVEL IS ATOM

GGCAT IS MCY UNS AT 35

DEFAULT ECLEVEL IS LIMITED

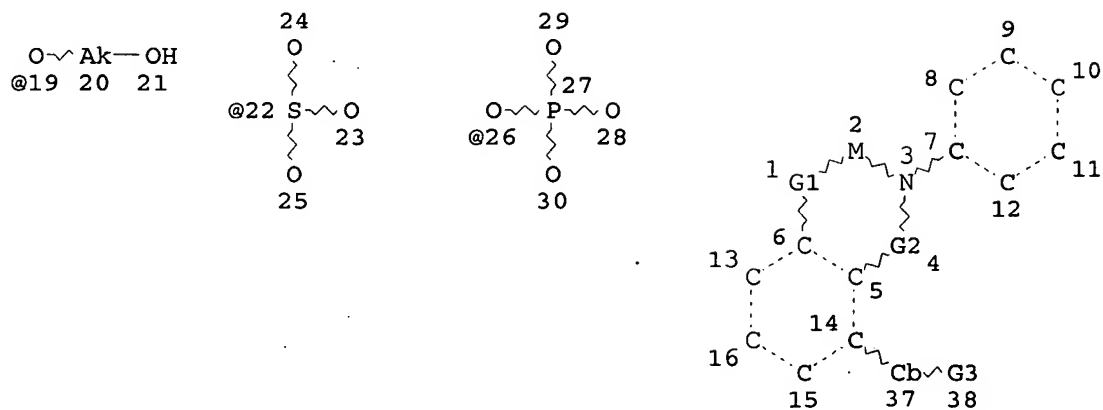
GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 31

STEREO ATTRIBUTES: NONE

L23 STR



@39 N +

VAR G1=O/S/N

VAR G2=C/N

VAR G3=19/22/26/39

NODE ATTRIBUTES:

CHARGE IS *+ AT 39

DEFAULT MLEVEL IS ATOM

GGCAT IS MCY UNS AT 37

DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 31

STEREO ATTRIBUTES: NONE

L25 0 SEA FILE=REGISTRY SUB=L7 SSS FUL (L19 OR L20 OR L21 OR
L22 OR L23)

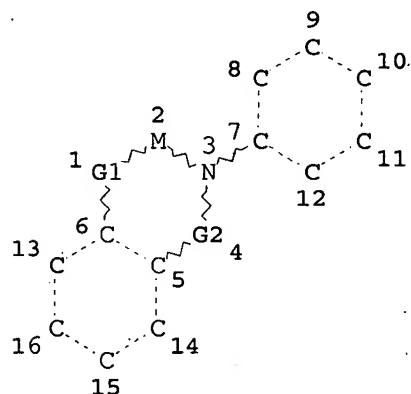
100.0% PROCESSED 10076 ITERATIONS

0 ANSWERS

SEARCH TIME: 00.00.01

=> d que stat l34

L3 STR



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VAR G2=C/N

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM

DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

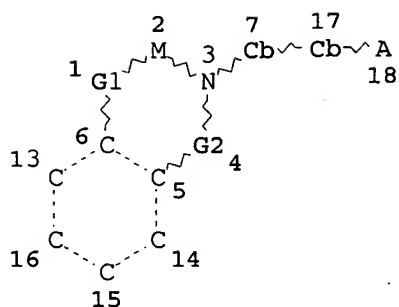
RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 16

STEREO ATTRIBUTES: NONE

L7 10084 SEA FILE=REGISTRY SSS FUL L3

L26 STR



VAR G1=O/S/N

VAR G2=C/N

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM

GGCAT IS MCY UNS AT 7

GGCAT IS MCY UNS AT 17

DEFAULT ECLÉVEL IS LIMITED

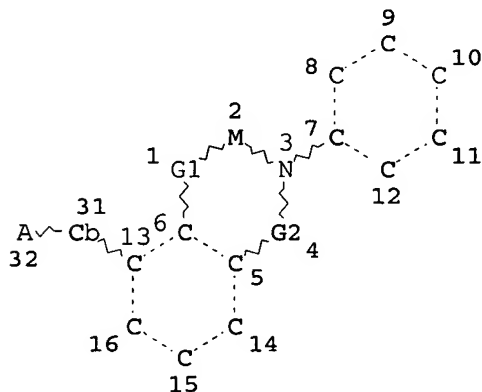
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RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 13

STEREO ATTRIBUTES: NONE

L27 STR



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NODE ATTRIBUTES:

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GGCAT IS MCY UNS AT 31

DEFAULT ECLEVEL IS LIMITED

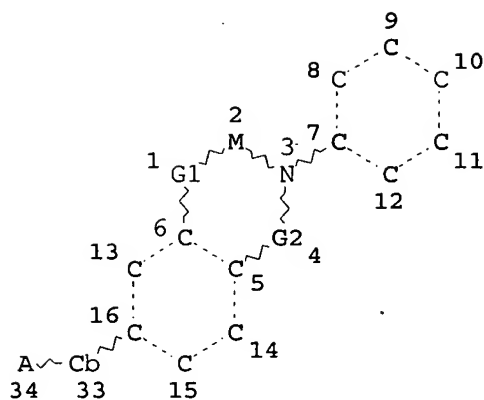
GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 18

STEREO ATTRIBUTES: NONE

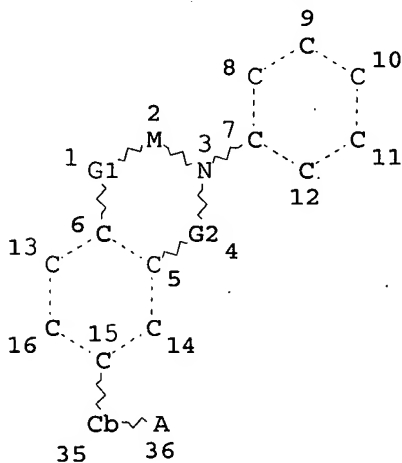
L28 STR



VAR G1=O/S/N
 VAR G2=C/N
 NODE ATTRIBUTES:
 DEFAULT MLEVEL IS ATOM
 GGCAT IS MCY UNS AT 33
 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
 RING(S) ARE ISOLATED OR EMBEDDED
 NUMBER OF NODES IS 18

STEREO ATTRIBUTES: NONE
 L29 STR



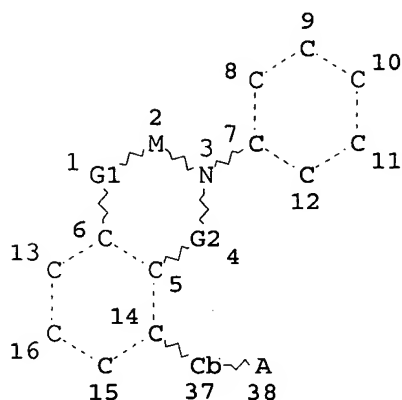
VAR G1=O/S/N
 VAR G2=C/N
 NODE ATTRIBUTES:
 DEFAULT MLEVEL IS ATOM
 GGCAT IS MCY UNS AT 35
 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
 RING(S) ARE ISOLATED OR EMBEDDED
 NUMBER OF NODES IS 18

STEREO ATTRIBUTES: NONE

L30

STR



VAR G1=O/S/N

VAR G2=C/N

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM

GGCAT IS MCY UNS AT 37

DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 18

STEREO ATTRIBUTES: NONE

L32 303 SEA FILE=REGISTRY SUB=L7 SSS FUL (L26 OR L27 OR L28 OR L29 OR L30)

L33 299 SEA FILE=REGISTRY ABB=ON PLU=ON L32 AND M=1

L34 136 SEA FILE=REGISTRY ABB=ON PLU=ON L33 AND (B7 OR B8)/PG

=> fil hcap

FILE 'HCAPLUS' ENTERED AT 16:23:04 ON 13 DEC 2007

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FILE COVERS 1907 - 13 Dec 2007 VOL 147 ISS 25

FILE LAST UPDATED: 12 Dec 2007 (20071212/ED)

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=> d 140 ibib abs hitstr hitind 1-8

L40 ANSWER 1 OF 8 HCAPLUS COPYRIGHT 2007 ACS.on STN

ACCESSION NUMBER: 2007:128862 HCAPLUS

DOCUMENT NUMBER: 146:462351

TITLE: Nickel(II)-Methyl Complexes with Water-Soluble Ligands L [(salicylaldiminato- κ^2 N,O)NiMe(L)] and Their Catalytic Properties in Disperse Aqueous Systems

AUTHOR(S): Korthals, Brigitte; Goettker-Schnetmann, Inigo; Mecking, Stefan

CORPORATE SOURCE: Lehrstuhl fuer Chemische Materialwissenschaft, Fachbereich Chemie, Universitaet Konstanz, Konstanz, D-78457, Germany

SOURCE: Organometallics (2007), 26(6), 1311-1316
CODEN: ORGND7; ISSN: 0276-7333

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 146:462351

AB Neutral (salicylaldiminato)nickel(II) Me complexes $[\{6\text{-C(H):NAr-2,4-I2C6H2O-}\kappa^2\text{N,O}\}\text{NiMe(L)}]$ (Ar = 2,6-{3,5-(F3C)2C6H3}2C6H3) with different water-soluble ligands L (2a, L = 1,3,5-triaza-7-phosphaadamantane; 2b, L = hexamethylenetetramine (urotropine); 2c, L = tetraethylammonium pyridine-3-sulfonate; 2d, L = amino-terminated poly(ethylene glycol) monomethoxy ether) were prepared. 2A-d are potentially water-soluble catalyst precursors for ethylene polymerization, which form a water-insol. active site $[\{\kappa^2\text{-N,O}\}\text{NiR(ethylene)}]$ (R = growing chain). Only complex 2d was water-soluble (>2 mmol L⁻¹); 2c is soluble in water/2-propanol mixts. In toluene as a reaction medium, only the relatively weakly coordinated tertiary amine complex 2b is polymerization active ($1.7 + 104$ TO). In aqueous systems 2c,d are also active due to compartmentalization of the active site in the polymer particles and of L in the aqueous phase. Polyethylene particle sizes vary from 18 nm (dispersions formed with 2d) to over 0.5 μm (2c) to suspensions (2b) depending on the initial state of the reaction mixture, correlated with catalyst solubility.

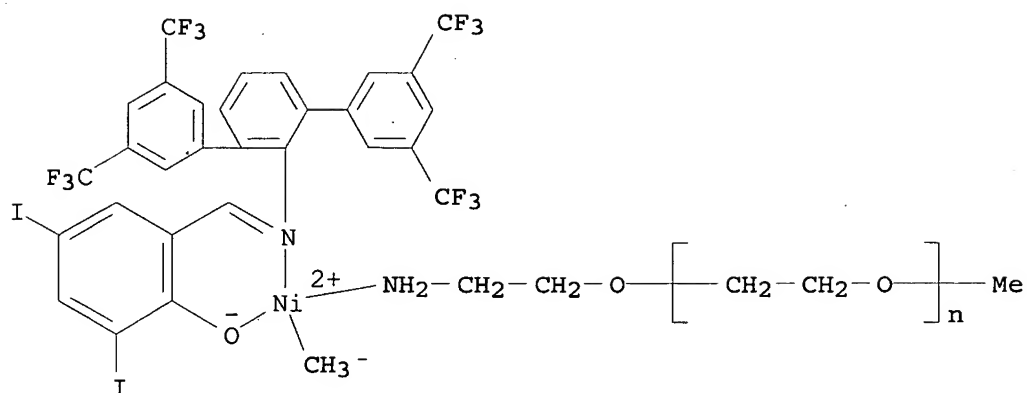
IT 900531-15-9

RL: CAT (Catalyst use); USES (Uses)

(preparation and ethylene polymerization catalytic activity of salicylaldiminato nickel Me complexes with water-soluble ligands)

RN 900531-15-9 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α -methyl- ω -hydroxy-, ether with [2-(amino- κ N)ethanol][2,4-diiodo-6-[[[3,3',5,5'-tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino- κ N]methyl]phenolato- κ O]nickel (1:1) (CA INDEX NAME)



IT 934843-93-3P 934843-94-4P 934843-96-6P

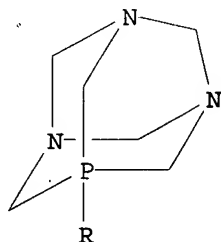
RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(preparation and ethylene polymerization catalytic activity of salicylaldiminato nickel Me complexes with water-soluble ligands)

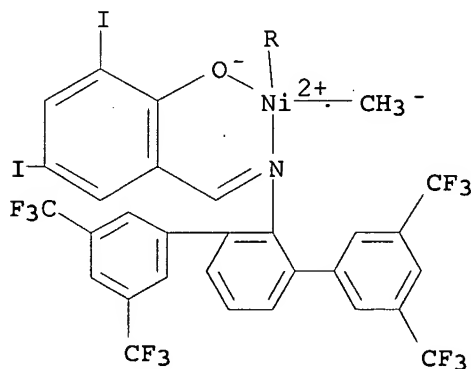
RN 934843-93-3 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl) [1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]methyl (1,3,5-triaza-7-phosphatricyclo[3.3.1.1.3,7]decane-κP7)-, (SP-4-3)- (CA INDEX NAME)

PAGE 1-A

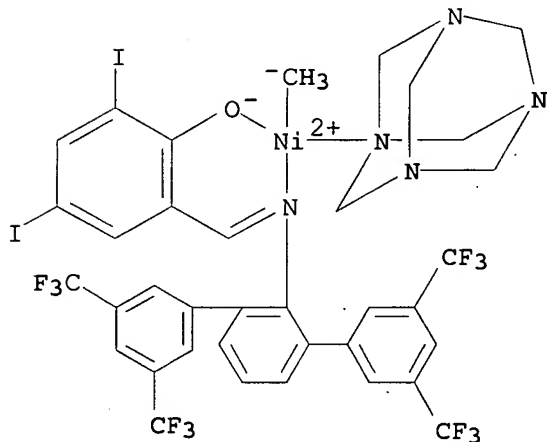


PAGE 2-A



RN 934843-94-4 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl) [1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]methyl (1,3,5,7-tetraazatricyclo[3.3.1.1^{3,7}]decane-κN1)-, (SP-4-4)- (CA INDEX NAME)



RN 934843-96-6 HCAPLUS

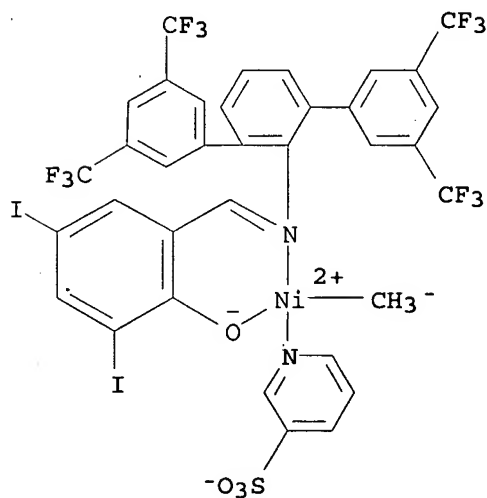
CN Ethanaminium, N,N,N-triethyl-, (SP-4-4)-[2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl) [1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]methyl (3-pyridinesulfonato-κN1)nickelate(1-) (1:1) (CA INDEX NAME)

CM 1

CRN 934843-95-5

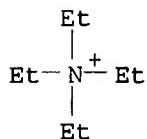
CMF C35 H19 F12 I2 N2 Ni O4 S

CCI CCS



CM 2

CRN 66-40-0
CMF C8 H20 N



CC 29-13 (Organometallic and Organometalloidal Compounds)
Section cross-reference(s): 35

ST salicylaldiminato nickel methyl water soluble ligand complex prepn
catalyst; **polymn** catalyst salicylaldiminato nickel methyl
water soluble ligand complex

IT **Polymerization**
Polymerization catalysts
(preparation and ethylene **polymerization** catalytic activity of
salicylaldiminato nickel Me complexes with water-soluble ligands)

IT 900531-15-9
RL: CAT (Catalyst use); USES (Uses)
(preparation and ethylene **polymerization** catalytic activity of
salicylaldiminato nickel Me complexes with water-soluble ligands)

IT 934843-93-3P 934843-94-4P 934843-96-6P
RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP
(Preparation); USES (Uses)
(preparation and ethylene **polymerization** catalytic activity of
salicylaldiminato nickel Me complexes with water-soluble ligands)

IT 74-85-1, Ethylene, reactions 100-97-0, Urotropine, reactions
636-73-7, 3-Pyridinesulfonic acid 53597-69-6, 1,3,5-Triaza-7-
phosphaadamantane 122905-76-4 667938-70-7
RL: RCT (Reactant); RACT (Reactant or reagent)
(preparation and ethylene **polymerization** catalytic activity of
salicylaldiminato nickel Me complexes with water-soluble ligands)

IT 934843-92-2P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);
RACT (Reactant or reagent)
(preparation and ethylene **polymerization** catalytic activity of
salicylaldiminato nickel Me complexes with water-soluble ligands)

IT 76-59-5, Bromothymol blue
RL: RGT (Reagent); RACT (Reactant or reagent)
(preparation and ethylene **polymerization** catalytic activity of
salicylaldiminato nickel Me complexes with water-soluble ligands)

IT 9002-88-4P, Polyethylene
RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation and ethylene **polymerization** catalytic activity of
salicylaldiminato nickel Me complexes with water-soluble ligands)

REFERENCE COUNT: 47 THERE ARE 47 CITED REFERENCES AVAILABLE
FOR THIS RECORD. ALL CITATIONS AVAILABLE
IN THE RE FORMAT

L40 ANSWER 2 OF 8 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2006:766770 HCAPLUS

DOCUMENT NUMBER: 145:377582

TITLE: **Aqueous Dispersions of**
Polypropylene and Poly(1-butene) with Variable
Microstructures Formed with Neutral Nickel(II)
Complexes

AUTHOR(S): Wehrmann, Peter; Mecking, Stefan

CORPORATE SOURCE: Fachbereich Chemie, Universitaet Konstanz,
Konstanz, D-78457, Germany
SOURCE: Macromolecules (2006), 39(18), 5963-5964
CODEN: MAMOBX; ISSN: 0024-9297
PUBLISHER: American Chemical Society
DOCUMENT TYPE: Journal
LANGUAGE: English

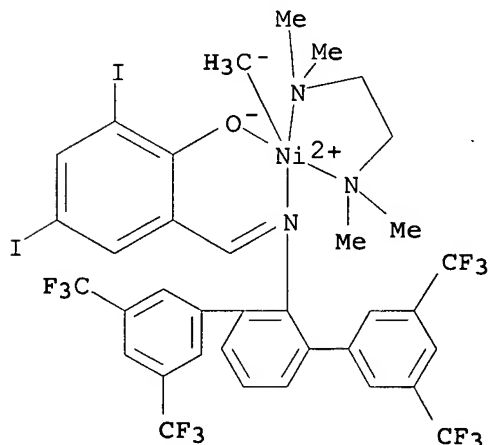
AB **Polymerization of 1-olefins in the presence of neutral Ni(II) complexes in aqueous emulsions gave poly(1-olefin) dispersions.** Microstructure analyses reveal that insertion occurs in primary alkyls exclusively, formed by 1,2- and 2,1-insertion, and subsequent chain running. 2,1-Insertion was unambiguously shown for a neutral Ni(II) **polymerization catalyst** for the first time. Microstructures can be varied by control of the chain running capacity of the catalyst via remote substituents of the salicylaldimine ligand.

IT 869902-26-1 910919-40-3 910919-41-4

RL: CAT (Catalyst use); USES (Uses)
(aqueous dispersions of polypropylene and polybutene with variable microstructures formed with neutral nickel complexes)

RN 869902-26-1 HCAPLUS

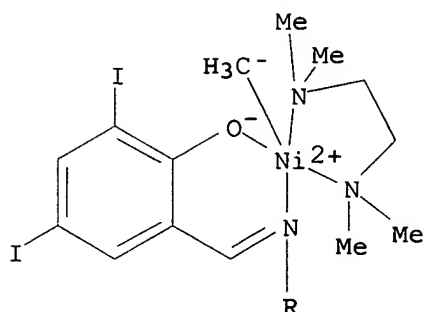
CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]methyl(N,N,N',N'-tetramethyl-1,2-ethanediamine-κN,κN')- (9CI) (CA INDEX NAME)



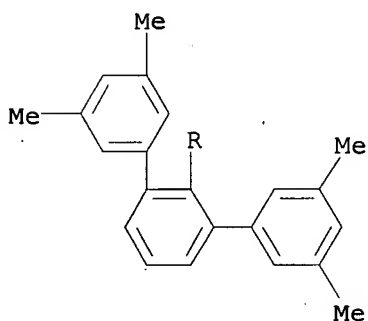
RN 910919-40-3 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetramethyl[1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]methyl(N,N,N',N'-tetramethyl-1,2-ethanediamine-κN,κN')- (9CI) (CA INDEX NAME)

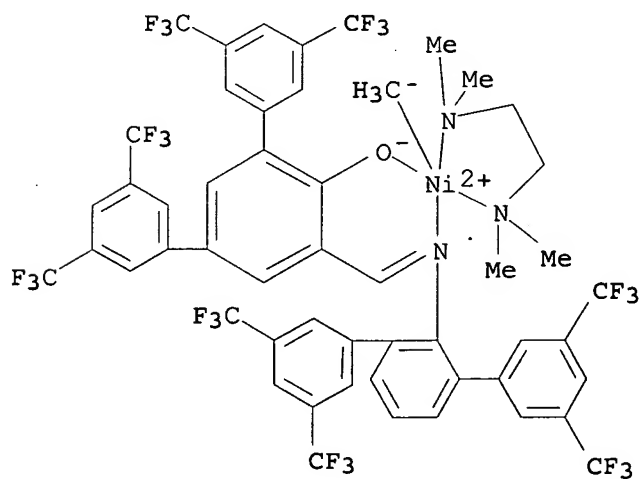
PAGE 1-A



PAGE 2-A



RN 910919-41-4 HCAPLUS
 CN Nickel, methyl[5'-[[[3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-κN)methyl]-3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-4'-olato-κO](N,N,N',N'-tetramethyl-1,2-ethanediamine-κN,κN')-(9CI) (CA INDEX NAME)



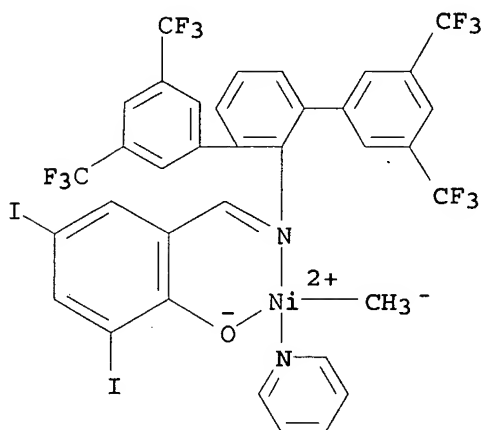
CC 35-3 (Chemistry of Synthetic High Polymers)
 ST nickel complex catalyst polymn polypropylene polybutene

L40 ANSWER 3 OF 8 HCAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER: 2006:766765 HCAPLUS
DOCUMENT NUMBER: 145:377603
TITLE: Copolymerization of Ethylene with 1-Butene and
Norbornene to Higher Molecular Weight Copolymers
in **Aqueous Emulsion**
AUTHOR(S): Wehrmann, Peter; Zuideveld, Martin; Thomann,
Ralf; Mecking, Stefan
CORPORATE SOURCE: Fachbereich Chemie, Universitaet Konstanz,
Konstanz, D-78457, Germany
SOURCE: Macromolecules (2006), 39(18), 5995-6002
CODEN: MAMOBX; ISSN: 0024-9297
PUBLISHER: American Chemical Society
DOCUMENT TYPE: Journal
LANGUAGE: English

IT 680185-26-6 680185-29-9
RL: CAT (Catalyst use); USES (Uses)
(polymerization catalysts; copolymn. of ethylene with 1-butene
and norbornene to higher mol. weight copolymers in aqueous
emulsion)

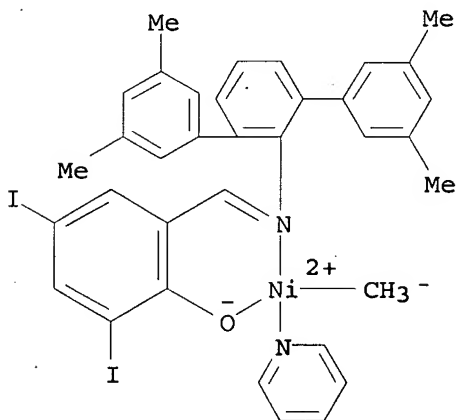
12/13/2007

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl) [1,1':3',1''-terphenyl]-2'-yl]imino-κN)methyl]phenolato-κO)methyl(pyridine)-, (SP-4-4)- (9CI) (CA INDEX NAME)



RN 680185-29-9 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetramethyl [1,1':3',1''-terphenyl]-2'-yl]imino-κN)methyl]phenolato-κO)methyl(pyridine)-, (SP-4-4)- (9CI) (CA INDEX NAME)



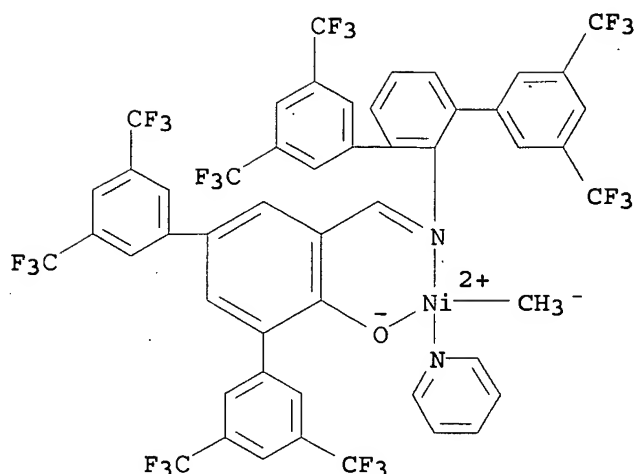
IT 910858-45-6P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(polymerization catalysts; copolymn. of ethylene with 1-butene and norbornene to higher mol. weight copolymers in aqueous emulsion)

RN 910858-45-6 HCAPLUS

CN Nickel, methyl(pyridine) [5'-[[[3,3'',5,5''-tetrakis(trifluoromethyl) [1,1':3',1''-terphenyl]-2'-yl]imino-κN)methyl]-3,3'',5,5''-tetrakis(trifluoromethyl) [1,1':3',1''-terphenyl]-4'-olato-κO]-, (SP-4-4)- (9CI) (CA INDEX NAME)



- CC 35-4 (Chemistry of Synthetic High Polymers)
Section cross-reference(s): 29, 36
- IT Crystallinity
Melting point
Molecular weight
Molecular weight distribution
Polymer morphology
(characterization of ethylene copolymers with 1-butene and norbornene made in **aqueous emulsion**)
- IT **Polymerization catalysts**
(coordination complex; in copolymn. of ethylene with 1-butene and norbornene to higher mol. weight copolymers in **aqueous emulsion**)
- IT **Latex**
(copolymn. of ethylene with 1-butene and norbornene to higher mol. weight copolymers in **aqueous emulsion**)
- IT 25087-34-7P, 1-Butene-ethylene copolymer 26007-43-2P,
Ethylene-norbornene copolymer
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
(copolymn. of ethylene with 1-butene and norbornene to higher mol. weight copolymers in **aqueous emulsion**)
- IT 2631-77-8, 3,5-Diiodosalicylaldehyde 73852-19-4,
3,5-Bis(trifluoromethyl)phenylboronic acid 122905-76-4,
Dimethyl(N,N,N',N'-tetramethylethylenediamine)nickel 667938-69-4,
2,6-Bis(3,5-bis(trifluoromethyl)phenyl)aniline
RL: RCT (Reactant); RACT (Reactant or reagent)
(copolymn. of ethylene with 1-butene and norbornene to higher mol. weight copolymers in **aqueous emulsion**)
- IT 910858-46-7P 910858-47-8P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(copolymn. of ethylene with 1-butene and norbornene to higher mol. weight copolymers in **aqueous emulsion**)
- IT 332951-99-2 680185-26-6 680185-29-9
RL: CAT (Catalyst use); USES (Uses)
(**polymerization catalysts**; copolymn. of ethylene with 1-butene and norbornene to higher mol. weight copolymers in **aqueous emulsion**)
- IT 910858-45-6P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP
 (Preparation); USES (Uses)
 (polymerization catalysts; copolymn. of ethylene with 1-butene
 and norbornene to higher mol. weight copolymers in aqueous
 emulsion)

REFERENCE COUNT: 42 THERE ARE 42 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L40 ANSWER 4 OF 8 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2006:495729 HCAPLUS

DOCUMENT NUMBER: 145:167619

TITLE: Water-Soluble Salicylaldiminato Ni(II)-Methyl
 Complexes: Enhanced Dissociative Activation for
 Ethylene Polymerization with
 Unprecedented Nanoparticle Formation

AUTHOR(S): Goettker-Schnetmann, Inigo; Korthals, Brigitte;
 Mecking, Stefan

CORPORATE SOURCE: Fachbereich Chemie, Universitaet Konstanz,
 Konstanz, D-78457, Germany

SOURCE: Journal of the American Chemical Society (2006),
 128(24), 7708-7709

CODEN: JACSAT; ISSN: 0002-7863

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 145:167619

AB Water-soluble single-component (κ^2 -N,O)-salicylaldiminato Ni-Me
 complexes having ligands tri(sodium phenylsulfonate)phosphine,
 di(sodium phenylsulfonate)phenylphosphine (I), and H₂N(CH₂CH₂O)_nMe
 (II) catalyzed the polymerization of ethylene under organic
 solvent-free aqueous conditions producing high-mol.-weight
 polyethylene with particle sizes <10 nm. Amphiphilic I- and
 II-complexes exhibited a polymerization activity in water higher
 than that in toluene. A solvation-favored activation of
 precatalysts by equilibrium dissociation of ligands in aqueous solution
 likely accounted for this enhanced polymerization activity. The
 observed generation of a given particle by a single active site was an
 unprecedented mechanism for formation of aqueous particle
 dispersions.

IT 900531-09-1P 900531-10-4P 900531-11-5P

900531-12-6P 900531-13-7P 900531-14-8P

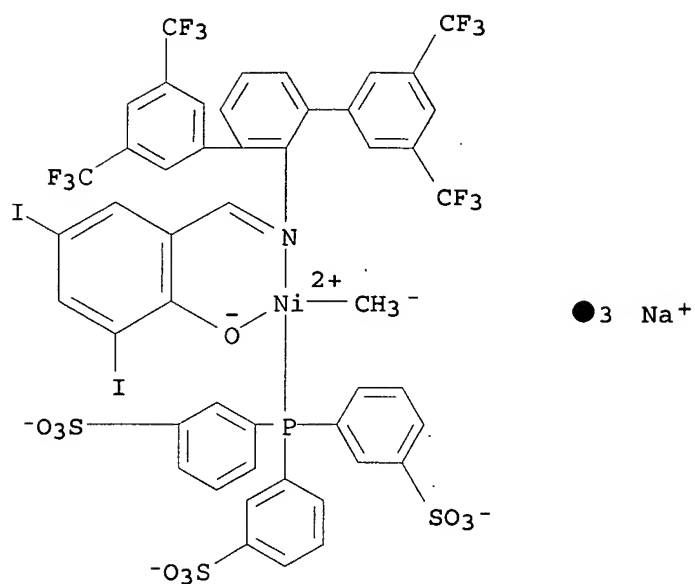
900531-15-9P 900531-16-0P 900531-17-1P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP
 (Preparation); USES (Uses)

(salicylaldiminatomethylnickel complexes for ethylene
 polymerization with nanoparticle formation)

RN 900531-09-1 HCAPLUS

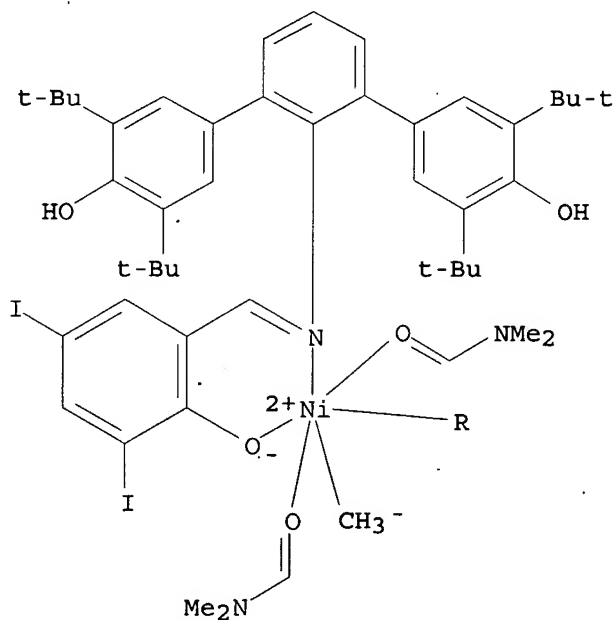
CN Nickelate(3-), [2,4-diiodo-6-[[[3,3'',5,5''-
 tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-
 κ N)methyl]phenolato- κ O)methyl[[3,3',3''-(phosphinidyne-
 κ P)tris[benzenesulfonato]](3-)]-, trisodium (9CI) (CA INDEX
 NAME)



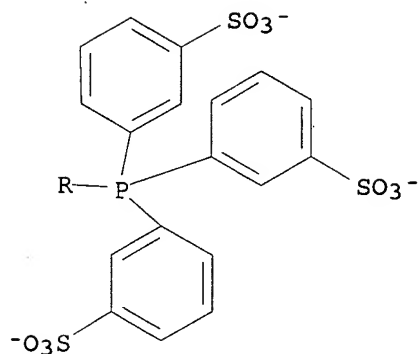
RN 900531-10-4 HCAPLUS

CN Nickelate(3-), bis(N,N-dimethylformamide-κO)methyl[[3,3',3''-(phosphinidyne-κP)tris[benzenesulfonato]](3-)][3,3'',5,5''-tetrakis(1,1-dimethylethyl)-2'-[[[2-(hydroxy-κO)-3,5-diiodophenyl]methylene]amino-κN][1,1':3',1''-terphenyl]-4,4''-diolato]-, trisodium (9CI) (CA INDEX NAME)

PAGE 1-A

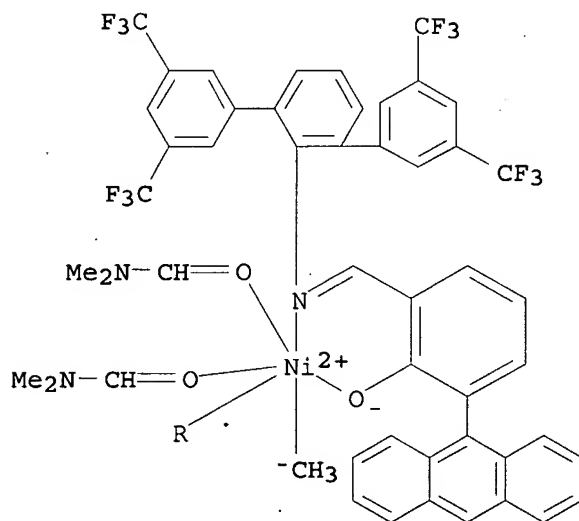


PAGE 2-A

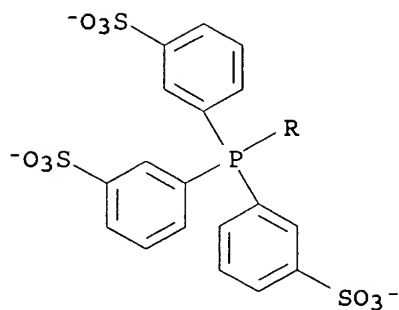
● 3 Na⁺

RN 900531-11-5 HCAPLUS
 CN Nickelate(3-), [2-(9-anthracenyl)-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]bis(N,N-dimethylformamide-κO)methyl[[3,3',3''-(phosphinidyne-κP)tris[benzenesulfonato]](3-)]-, trisodium (9CI) (CA INDEX NAME)

PAGE 1-A

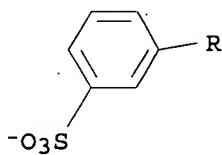
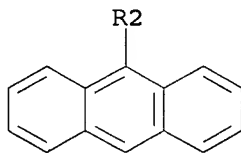


PAGE 2-A

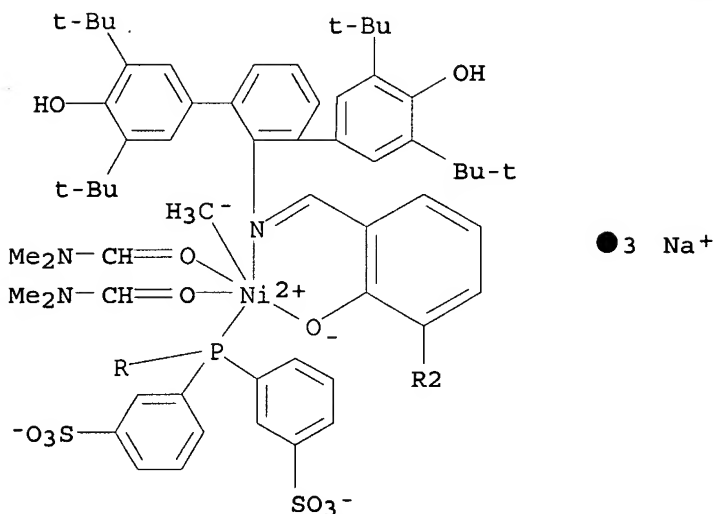
●3 Na⁺

RN 900531-12-6 HCAPLUS
 CN Nickelate(3-), [2'-[[[3-(9-anthracenyl)-2-(hydroxy-
 κ O)phenyl]methylene]amino- κ N]-3,3'',5,5''-tetrakis(1,1-
 dimethylethyl)[1,1':3',1''-terphenyl]-4,4''-diolato]bis(N,N-
 dimethylformamide- κ O)methyl[[3,3',3''-(phosphinidyne-
 κ P)tris[benzenesulfonato]](2-)]-, trisodium (9CI) (CA INDEX
 NAME)

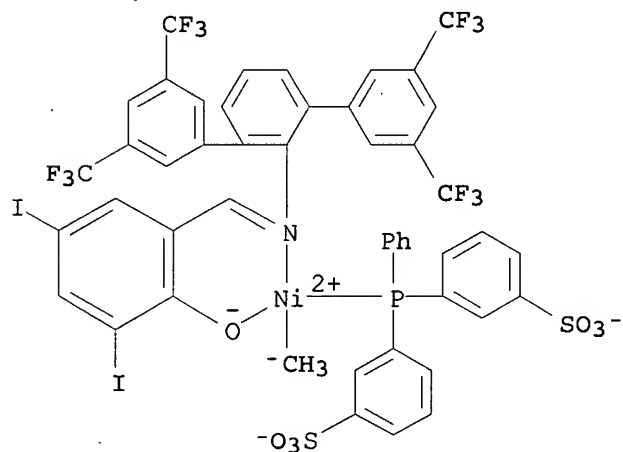
PAGE 1-A



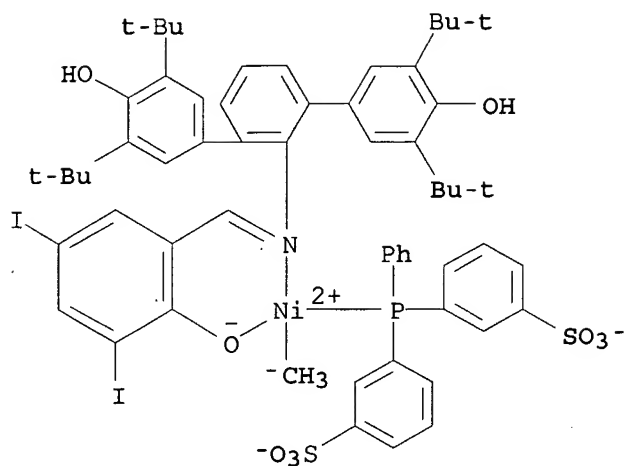
PAGE 2-A



RN 900531-13-7 HCAPLUS
 CN Nickelate(2-), [2,4-diiodo-6-[[[3,3',5,5'-tetrakis(trifluoromethyl)1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]methyl[[3,3'-(phenylphosphinidene-κP)bis[benzenesulfonato]](2-)]-, disodium (9CI) (CA INDEX NAME)

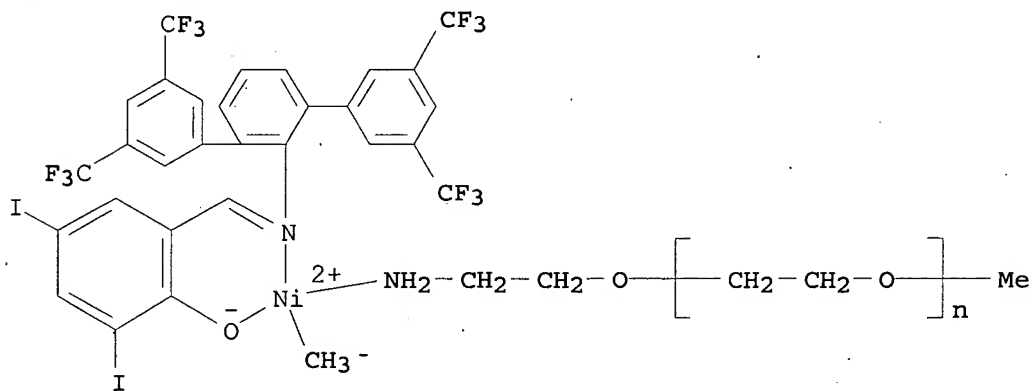


RN 900531-14-8 HCAPLUS
 CN Nickelate(2-), methyl[[3,3'-(phenylphosphinidene-κP)bis[benzenesulfonato]](2-)][3,3',5,5'-tetrakis(1,1-dimethylethyl)-2'-[[[2-(hydroxy-κO)-3,5-diiodophenyl]methylene]amino-κN][1,1':3',1''-terphenyl]-4,4''-diolato]-, disodium (9CI) (CA INDEX NAME)

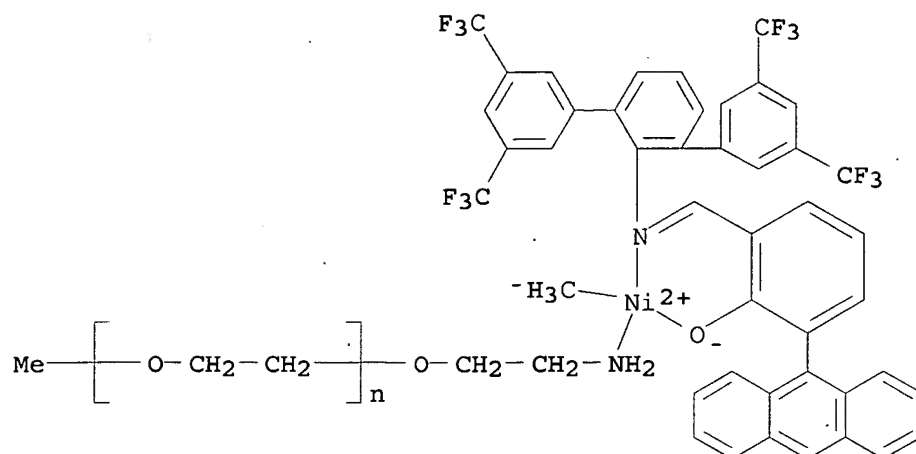


● 2 Na⁺

RN 900531-15-9 HCAPLUS
 CN Poly(oxy-1,2-ethanediyl), α-methyl-ω-hydroxy-, ether
 with [2-(amino-κN)ethanol] [2,4-diiodo-6-[[[3,3',5,5'-
 tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-
 κN]methyl]phenolato-κO]nickel (1:1) (CA INDEX NAME)

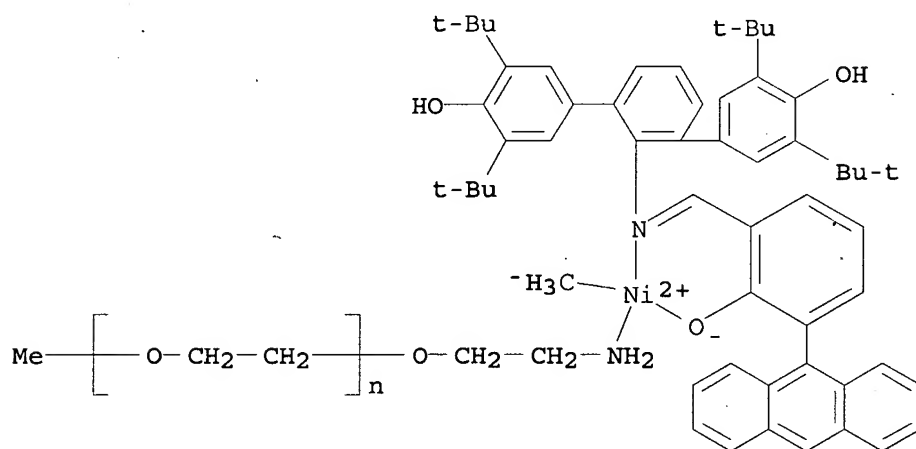


RN 900531-16-0 HCAPLUS
 CN Poly(oxy-1,2-ethanediyl), α-methyl-ω-hydroxy-, ether
 with [2-(amino-κN)ethanol] [2-(9-anthracenyl)-6-[[[3,3',5,5'-
 tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-
 κN]methyl]phenolato-κO]methylnickel (9CI) (CA INDEX
 NAME)



RN 900531-17-1 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α -methyl- ω -hydroxy-, ether with [2-(amino- κ N)ethanol][2'-[[[3-(9-anthracenyl)-2-(hydroxy- κ O)phenyl]methylene]amino- κ N]-3,3',5,5'-tetrakis(1,1-dimethylethyl)[1,1':3',1''-terphenyl]-4,4''-diolato]methylnickel (1:1) (9CI) (CA INDEX NAME)



CC 35-4 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 29, 67

ST ethylene polymn catalyst salicylaldiminato nickel methyl complex

IT Ligands

RL: CAT (Catalyst use); USES (Uses)

(nickel complexes; salicylaldiminatomethylnickel complexes for ethylene polymerization with nanoparticle formation)

IT Dispersion (of materials)

Nanoparticles

Polymerization catalysts

(salicylaldiminatomethylnickel complexes for ethylene polymerization with nanoparticle formation)

IT Coordination compounds

RL: CAT (Catalyst use); USES (Uses)

(salicylaldiminatomethylnickel complexes for ethylene polymerization with nanoparticle formation)

IT 900531-09-1P 900531-10-4P 900531-11-5P
900531-12-6P 900531-13-7P 900531-14-8P
900531-15-9P 900531-16-0P 900531-17-1P
RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(salicylaldiminatomethylnickel complexes for ethylene polymerization with nanoparticle formation)

IT 9002-88-4P, Polyethylene
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(salicylaldiminatomethylnickel complexes for ethylene polymerization with nanoparticle formation)

IT 74-85-1, Ethylene, reactions 2631-77-8, 2-Hydroxy-3,5-diiodobenzaldehyde 63995-70-0, TPPTS 64018-22-0, TPPDS 80506-64-5 122905-76-4 210096-14-3 667938-69-4 900531-18-2
RL: RCT (Reactant); RACT (Reactant or reagent)

(salicylaldiminatomethylnickel complexes for ethylene polymerization with nanoparticle formation)

IT 900531-19-3P 900531-20-6P 900531-21-7P 935656-32-9P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(salicylaldiminatomethylnickel complexes for ethylene polymerization with nanoparticle formation)

REFERENCE COUNT: 22 THERE ARE 22 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L40 ANSWER 5 OF 8 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2005:1059417 HCAPLUS

DOCUMENT NUMBER: 144:23148

TITLE: Possible Side Reactions Due to Water in

Emulsion Polymerization by

Late Transition Metal Complexes. 1. Water

Complexation and Hydrolysis of the Growing Chain

AUTHOR(S): Hristov, Iordan H.; DeKock, Roger L.; Anderson, Grant D. W.; Goettker-Schnetmann, Inigo; Mecking, Stefan; Ziegler, Tom

CORPORATE SOURCE: Department of Chemistry, University of Calgary, Calgary, AB, T2N 1N4, Can.

SOURCE: Inorganic Chemistry (2005), 44(22), 7806-7818
CODEN: INOCAJ; ISSN: 0020-1669

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

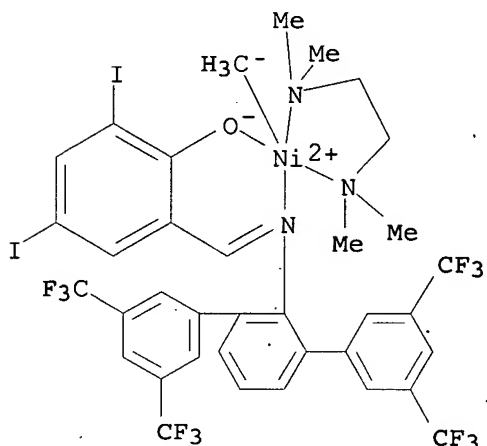
AB The transition metal catalyzed ethylene polymerization in aqueous emulsion has been increasingly successful in the last couple of years. Water however adversely affects the polymerization process by (a) competing with ethylene for the binding site at the metal and (b) hydrolyzing the growing chain. Neutral salicylaldiminato and cationic diimine complexes of Ni and Pd with different substituent patterns are studied here by a functional theory to determine their propensity toward water complexation and hydrolysis of the growing chain. Exptl. NMR studies have also been carried out on the protonolysis of the Ni(II)-based Grubbs catalyst. It is found that in general that (a) ethylene coordination is preferred over water coordination for both Ni and Pd catalysts and (b) hydrolysis of the metal alkyl bond is competitive to ethylene insertion.

IT 869902-26-1

RL: RCT (Reactant); RACT (Reactant or reagent)
 (protonolysis of; water complexation and hydrolysis of the
 growing polyethylene chain as possible side reactions in ethylene
polymerization in aqueous emulsions with Grubbs
 and Brookhart catalyst systems)

RN 869902-26-1 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':
 3',1''-terphenyl]-2'-yl]imino-κN)methyl]phenolato-
 κO)methyl(N,N,N',N'-tetramethyl-1,2-ethanediamine-
 κN,κN')- (9CI) (CA INDEX NAME)



CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 29, 78

ST nickel palladium salicylaldiminato diimine complex ethylene water
 coordination competition; ethylene **polymn aq**
emulsion side reaction Grubbs Brookhart catalyst; quantum
 mech calcn complexation energy water ethylene **polymn**

IT **Polymerization**

(**emulsion**; water complexation and hydrolysis of the
 growing polyethylene chain as possible side reactions in ethylene
polymerization in aqueous emulsions with Grubbs
 and Brookhart catalyst systems)

IT Complexation enthalpy

Density functional theory

Hydrolysis

Polymerization catalysts

Solvent effect

Transition state structure

(water complexation and hydrolysis of the growing polyethylene
 chain as possible side reactions in ethylene **polymerization in**
aqueous emulsions with Grubbs and Brookhart
 catalyst systems)

IT Transition metal complexes

RL: CAT (Catalyst use); CPS (Chemical process); PEP (Physical,
 engineering or chemical process); PRP (Properties); PROC (Process);
 USES (Uses)

(water complexation and hydrolysis of the growing polyethylene
 chain as possible side reactions in ethylene **polymerization in**
aqueous emulsions with Grubbs and Brookhart
 catalyst systems)

IT 869902-26-1 870481-46-2
RL: RCT (Reactant); RACT (Reactant or reagent)
(protonolysis of; water complexation and hydrolysis of the
growing polyethylene chain as possible side reactions in ethylene
polymerization in aqueous emulsions with Grubbs
and Brookhart catalyst systems)

IT 870481-11-1
RL: CAT (Catalyst use); CPS (Chemical process); PEP (Physical,
engineering or chemical process); PRP (Properties); PROC (Process);
USES (Uses)
(water complexation and hydrolysis of the growing polyethylene
chain as possible side reactions in ethylene polymerization in
aqueous emulsions with Grubbs and Brookhart
catalyst systems)

IT 181708-22-5 870481-12-2 870481-13-3 870481-14-4 870481-15-5
870481-16-6 870481-17-7 870481-18-8 870481-19-9 870481-20-2
870481-21-3 870481-22-4 870481-23-5 870481-24-6 870481-25-7
870481-26-8 870481-27-9 870481-28-0 870481-29-1 870481-30-4
870481-31-5 870481-32-6 870481-33-7 870481-34-8 870481-35-9
870481-36-0 870481-37-1 870481-38-2 870481-39-3 870481-40-6
870481-41-7 870481-42-8 870481-43-9 870481-44-0 870481-45-1
RL: CAT (Catalyst use); PRP (Properties); USES (Uses)
(water complexation and hydrolysis of the growing polyethylene
chain as possible side reactions in ethylene polymerization in
aqueous emulsions with Grubbs and Brookhart
catalyst systems)

IT 74-85-1, Ethylene, processes 7732-18-5, Water, processes
RL: CPS (Chemical process); PEP (Physical, engineering or chemical
process); PROC (Process)
(water complexation and hydrolysis of the growing polyethylene
chain as possible side reactions in ethylene polymerization in
aqueous emulsions with Grubbs and Brookhart
catalyst systems)

IT 9002-88-4P, Polyethylene
RL: SPN (Synthetic preparation); PREP (Preparation)
(water complexation and hydrolysis of the growing polyethylene
chain as possible side reactions in ethylene polymerization in
aqueous emulsions with Grubbs and Brookhart
catalyst systems)

REFERENCE COUNT: 85 THERE ARE 85 CITED REFERENCES AVAILABLE
FOR THIS RECORD. ALL CITATIONS AVAILABLE
IN THE RE FORMAT

L40 ANSWER 6 OF 8 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2005:1039667 HCAPLUS

DOCUMENT NUMBER: 144:7136

TITLE: A General Route to Very Small Polymer Particles
with Controlled Microstructures

AUTHOR(S): Monteil, Vincent; Wehrmann, Peter; Mecking,
Stefan

CORPORATE SOURCE: Fachbereich Chemie, Universitaet Konstanz,
Konstanz, D-78457, Germany

SOURCE: Journal of the American Chemical Society (2005),
127(42), 14568-14569

CODEN: JACSAT; ISSN: 0002-7863

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 144:7136

AB A catalytic synthesis of previously inaccessible very small polymer

particles by means of catalyst microemulsion was reported. Known water-soluble lipophilic catalysts were employed. Very small polymer particles of 10-30 nm size with various microstructures (polyethylene, syndiotactic 1,2-polybutadiene, poly(cycloolefins)) are prepared by catalytic **polymerization** with **aqueous** catalyst microemulsion.

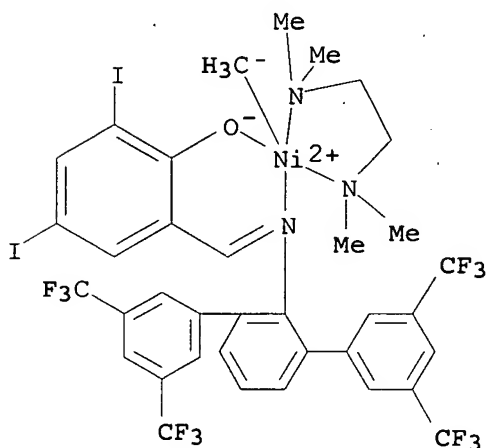
IT 869902-26-1

RL: CAT (Catalyst use); USES (Uses)

(general route to very small polymer particles with controlled microstructures via catalytic **polymerization** with **aq** catalyst microemulsion.)

RN 869902-26-1 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]methyl(N,N,N',N'-tetramethyl-1,2-ethanediamine-κN,κN')- (9CI) (CA INDEX NAME)



CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 36

ST polymer particle controlled microstructure; polycycloolefin small particle controlled microstructure; syndiotactic polybutadiene small particle controlled microstructure; polyethylene particle controlled microstructure catalyst microemulsion; **polymn** olefin water sol lipophilic catalyst

IT Surfactants

(anionic; general route to very small polymer particles with controlled microstructures via catalytic **polymerization** with **aqueous** catalyst microemulsion.)

IT Surfactants

(cationic; general route to very small polymer particles with controlled microstructures via catalytic **polymerization** with **aqueous** catalyst microemulsion.)

IT **Polymerization** catalysts

(**emulsion**, water soluble lipophilic; general route to very small polymer particles with controlled microstructures via catalytic **polymerization** with **aqueous** catalyst microemulsion.)

IT Molecular weight distribution

Particle size

Particle size distribution

(general route to very small polymer particles with controlled

- microstructures via catalytic **polymerization** with aq
. catalyst microemulsion.)
- IT Polyalkenamers
RL: SPN (Synthetic preparation); PREP (Preparation)
(general route to very small polymer particles with controlled
microstructures via catalytic **polymerization** with aq
. catalyst microemulsion.)
- IT Polymer chains
(microstructure; general route to very small polymer particles
with controlled microstructures via catalytic **polymerization**
with **aqueous** catalyst microemulsion.)
- IT 12154-10-8
RL: CAT (Catalyst use); USES (Uses)
(carbon sulfide system; general route to very small polymer
particles with controlled microstructures via catalytic
polymerization with **aqueous** catalyst microemulsion.)
- IT 3117-61-1, Salicyl aldimine
RL: CAT (Catalyst use); USES (Uses)
(catalyst precursor; general route to very small polymer
particles with controlled microstructures via catalytic
polymerization with **aqueous** catalyst microemulsion.)
- IT 118-75-2, Chloranil, reactions 603-35-0, Triphenylphosphine,
reactions 1295-35-8, Bis(1,5-Cyclooctadiene)nickel
RL: CAT (Catalyst use); RCT (Reactant); RACT (Reactant or reagent);
USES (Uses)
(catalyst precursor; general route to very small polymer
particles with controlled microstructures via catalytic
polymerization with **aqueous** catalyst microemulsion.)
- IT 75-15-0, Carbon sulfide (CS₂), uses
RL: CAT (Catalyst use); USES (Uses)
(cobalt complex system; general route to very small polymer
particles with controlled microstructures via catalytic
polymerization with **aqueous** catalyst microemulsion.)
- IT 172222-30-9 246047-72-3 869902-26-1
RL: CAT (Catalyst use); USES (Uses)
(general route to very small polymer particles with controlled
microstructures via catalytic **polymerization** with aq
. catalyst microemulsion.)
- IT 71-41-0, Pentanol, uses 151-21-3, SDS, surfactant, uses
1119-94-4, DTAB
RL: NUU (Other use, unclassified); USES (Uses)
(general route to very small polymer particles with controlled
microstructures via catalytic **polymerization** with aq
. catalyst microemulsion.)
- IT 9002-88-4P, Polyethylene 25038-76-0P, Polynorbornene 25267-51-0P
28702-45-6P, Poly(1-octene-1,8-diyl) 39366-06-8P 42813-64-9P,
Polynorbornene, sru
RL: SPN (Synthetic preparation); PREP (Preparation)
(general route to very small polymer particles with controlled
microstructures via catalytic **polymerization** with aq
. catalyst microemulsion.)
- IT 31567-90-5P, Syndiotactic polybutadiene
RL: SPN (Synthetic preparation); PREP (Preparation)
(of 1,2-configuration; general route to very small polymer
particles with controlled microstructures via catalytic
polymerization with **aqueous** catalyst microemulsion.)
- IT 7732-18-5, Water, uses
RL: NUU (Other use, unclassified); USES (Uses)
(toluene mixture solvent; general route to very small polymer
particles with controlled microstructures via catalytic

polymerization with aqueous catalyst microemulsion.)
IT 108-88-3, Toluene, uses
RL: NUU (Other use, unclassified); USES (Uses)
(water mixture solvent; general route to very small polymer
particles with controlled microstructures via catalytic
polymerization with aqueous catalyst microemulsion.)
REFERENCE COUNT: 24 THERE ARE 24 CITED REFERENCES AVAILABLE
FOR THIS RECORD. ALL CITATIONS AVAILABLE
IN THE RE FORMAT

L40 ANSWER 7 OF 8 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2004:231121 HCAPLUS

DOCUMENT NUMBER: 141:23933

TITLE: Catalytic ethylene polymerization in
aqueous emulsion: catalyst
tailoring and synthesis of very small latex
particles

AUTHOR(S): Bastero, Amaia; Kolb, Ludmila; Wehrmann, Peter;
Bauers, Florian; Goettker-Schnetmann, Inigo;
Monteil, Vincent; Thomann, Ralf; Chowdhry,
Mubarik; Mecking, Stefan

CORPORATE SOURCE: Freiburger Materialforschungszentrum und
Institut fuer Makromolekulare Chemie der
Albert-Ludwigs-Universitaet, Freiburg, D-79104,
Germany

SOURCE: PMSE Preprints (2004), 90, 740-741

CODEN: PPMRA9; ISSN: 1550-6703

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal; (computer optical disk)

LANGUAGE: English

AB Salicylaldiminato complexes [(N-O)Ni(Me)(py)] with N-terphenyl-R
substituted ligands display high catalytic activity in
polymerization of ethylene. Unexpectedly, the nature of the
substituent R has a dramatic effect on mol. weight, branching, and
crystallinity of prepared polyethylenes. While a semicryst. high mol.
weight polyethylene was obtained when R = CF₃, an electron-donating
group, e.g., R = Me led to low mol. weight, highly branched, completely
amorphous polyethylene. Reaction of tetrachlorobenzoquinone (TCBQ)
with water-soluble phosphines in water-miscible alc. solvent and
subsequent reaction with [Ni(cod)₂] affords a hydrophilic catalyst,
for emulsion polymerization without any addnl. liquid
organic phase. Stable polyethylene (mol. weight w ca. 10 g/mol) latexes
with up to 13% solids content were obtained, without optimization;
the latexes are transparent or only slightly hazy, indicating a very
small particle size.

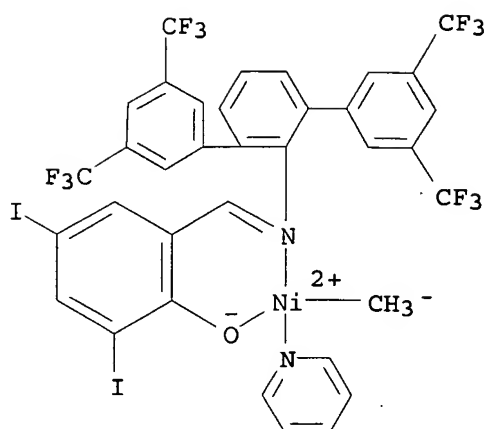
IT 680185-26-6

RL: CAT (Catalyst use); USES (Uses)

(polymerization catalyst; structure and substituent effects of
nickel salicylaldiminato catalysts in ethylene polymerization
to obtain very small latex particles)

RN 680185-26-6 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':
3',1''-terphenyl]-2'-yl]imino-κN)methyl]phenolato-
κO)methyl(pyridine)-, (SP-4-4)- (9CI) (CA INDEX NAME)



- CC 35-3 (Chemistry of Synthetic High Polymers)
Section cross-reference(s): 67, 78
- ST nickel salicylaldiminato terphenyl substituent effect catalyst
emulsion polymn ethylene; polyethylene latex
crystallinity mol wt nickel complex hydrophilic catalyst
- IT **Polymerization** catalysts
(emulsion; structure and substituent effects of nickel
salicylaldiminato catalysts in ethylene **polymerization** to
obtain very small latex particles)
- IT Substituent effects
(on catalyst activity; structure and substituent effects of
nickel salicylaldiminato catalysts in ethylene **polymerization**
to obtain very small latex particles)
- IT Crystallinity
Hydrophilicity
Latex
Particle size
(structure and substituent effects of nickel salicylaldiminato
catalysts in ethylene **polymerization** to obtain very small
latex particles)
- IT 118-75-2, Tetrachlorobenzoquinone, uses 603-35-0,
Triphenylphosphine, uses 1295-35-8, Bis(1,5-cyclooctadiene)nickel
RL: CAT (Catalyst use); USES (Uses)
(hydrophilic **polymerization** catalyst containing; structure and
substituent effects of nickel salicylaldiminato catalysts in
ethylene **polymerization** to obtain very small latex particles)
- IT 122905-76-4, Dimethyl(N,N,N',N'-tetramethylethylenediamine)nickel
RL: CAT (Catalyst use); USES (Uses)
(in-situ **polymerization** catalyst precursor; structure and
substituent effects of nickel salicylaldiminato catalysts in
ethylene **polymerization** to obtain very small latex particles)
- IT 680185-26-6
RL: CAT (Catalyst use); USES (Uses)
(**polymerization** catalyst; structure and substituent effects of
nickel salicylaldiminato catalysts in ethylene **polymerization**
to obtain very small latex particles)
- IT 9002-88-4P, Polyethylene
RL: SPN (Synthetic preparation); PREP (Preparation)
(structure and substituent effects of nickel salicylaldiminato
catalysts in ethylene **polymerization** to obtain very small
latex particles)

REFERENCE COUNT: 15 THERE ARE 15 CITED REFERENCES AVAILABLE

FOR THIS RECORD. ALL CITATIONS AVAILABLE
IN THE RE FORMAT

L40 ANSWER 8 OF 8 HCAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2004:198241 HCAPLUS
 DOCUMENT NUMBER: 140:236209
 TITLE: Procedure for the production of aqueous
 polymer dispersions by
 polymerization of olefins in the
 presence of transition metal complexes
 PATENT ASSIGNEE(S): BASF A.-G., Germany
 SOURCE: Ger. Offen., 18 pp.
 CODEN: GWXXBX
 DOCUMENT TYPE: Patent
 LANGUAGE: German
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 10240577	A1	20040311	DE 2002-10240577	200208 29
WO 2004020478	A1	20040311	WO 2003-EP8091	200307 24
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
AU 2003250152	A1	20040319	AU 2003-250152	200307 24
EP 1537150	A1	20050608	EP 2003-790811	200307 24
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK				
JP 2005536609	T	20051202	JP 2004-531814	200307 24
US 2005250920	A1	20051110	US 2005-524216	200502 10
PRIORITY APPLN. INFO.:				DE 2002-10240577 A
				200208 29
WO 2003-EP8091				W
				200307

24

OTHER SOURCE(S): MARPAT 140:236209

AB **Aqueous polymer dispersions** are manufactured by **polymerization** of olefins in the presence of complexes of Group 7-10 metals and azo or azomethine compds. having aromatic rings attach to both ends of the azo or azomethine group. A typical catalyst was manufactured by reaction of 2,6-bis[3,5-bis(trifluoromethyl)phenyl]aniline with 3,5-diiodo-2-hydroxybenzaldehyde and complexation of the resulting salicylaldimine ligand with tetramethylethylenediaminedime thylnickel.

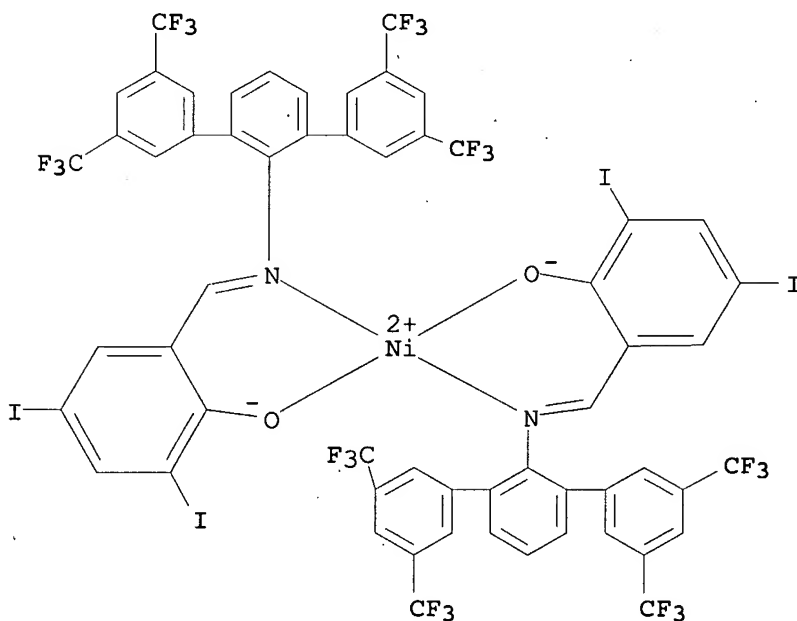
IT 667938-71-8P

RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)

RN 667938-71-8 HCAPLUS

CN Nickel, bis[2,4-diiodo-6-[[[3,3',5,5'-tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-κN]methyl]phenolato-κO]- (9CI) (CA INDEX NAME)



IC ICM C08F004-06

ICS C08F004-26; C08F002-16; C08F010-00

CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 67

ST transition metal arom azo deriv complex catalyst olefin **polymn**; azomethine arom deriv transition metal complex catalyst olefin **polymn**; salicylaldimine deriv nickel complex catalyst manuf olefin **polymn**

IT **Polymerization** catalysts

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)

IT Group VIIB. element complexes

Group VIII element complexes

RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)

IT Polyolefins

RL: IMF (Industrial manufacture); PREP (Preparation)

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)

IT Adhesives

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds. for adhesives)

IT Carpets

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds. for carpet treatment)

IT Leather

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds. for leather treatment)

IT Paints

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds. for paints)

IT Coating materials

Paper

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds. for paper coatings)

IT Pharmaceutical industry

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds. for pharmaceutical industry)

IT Plastic foams

RL: MSC (Miscellaneous)

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds. for plastic foams)

IT Textiles

(production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds. for textile treatment)

IT 87666-57-7P, 2,6-Diphenylaniline 667938-69-4P,

2,6-Bis[3,5-bis(trifluoromethyl)phenyl]aniline

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)

(catalyst ligand precursor; production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)

IT 98-80-6, Phenylboric acid 608-30-0, 2,6-Dibromoaniline

2631-77-8, 3,5-Diiodo-2-hydroxybenzaldehyde 274251-67-1,

3,5-Bis(trifluoromethyl)phenylboric acid

RL: RCT (Reactant); RACT (Reactant or reagent)

(catalyst ligand precursor; production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)

- IT 667938-70-7P
 RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (ligand; production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)
- IT 667938-71-8P
 RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses) (production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)
- IT 9002-88-4P, Polyethylene
 RL: IMF (Industrial manufacture); PREP (Preparation) (production of **aqueous polyolefin dispersions** by **polymerization** of olefins in presence of transition metal complexes of azo or azomethine compds.)

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only this hit structure was displayed.

L41 ANSWER 1 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2007:376022 HCAPLUS

DOCUMENT NUMBER: 147:10032

TITLE: Syntheses and Ethylene Polymerization Behavior of Supported Salicylaldimine-Based Neutral Nickel(II) Catalysts

AUTHOR(S): Hu, Tao; Li, Yan-Guo; Liu, Jing-Yu; Li, Yue-Sheng

CORPORATE SOURCE: State Key Laboratory of Polymer Physics and Chemistry, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, Changchun, 130022, Peop. Rep. China

SOURCE: Organometallics (2007), 26(10), 2609-2615
 CODEN: ORGND7; ISSN: 0276-7333

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 147:10032

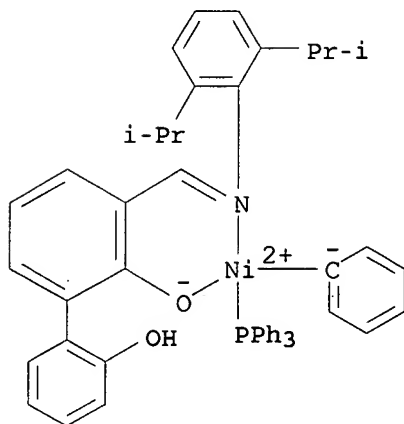
AB Two novel salicylaldimine-based neutral nickel(II) complexes, [(2,6-iPr₂C₆H₃)N:CH(2-ArC₆H₃O)]Ni(PPh₃)Ph (6, Ar = 2-(OH)C₆H₄; 8, Ar = 2-OH-3-(2,6-iPr₂C₆H₃N:CH)C₆H₃), have been synthesized, and their structures have also been confirmed by x-ray crystallog., elemental anal., and ¹H and ¹³C NMR spectra. An important structural feature of the two complexes is the free hydroxyl group, which allows them to react with silica pretreated with trimethylaluminum under immobilization by the formation of a covalent bond between the neutral nickel(II) complex and the pretreated silica. As active single-component catalysts, the two complexes exhibited high catalytic activities up to 1.14 and 1.47 + 10⁶ g PE/molNi·h for ethylene polymerization, resp., and yielded branched polymers. Requiring no cocatalyst, the two supported catalysts also showed relatively high activities up to 4.0 + 10⁵ g PE/molNi·h and produced polyethylenes with high weight-average mol. wts. of up to 120 kg/mol and a moderate degree of branching (ca. 13-26 branches per 1000 carbon atoms).

- IT 937808-87-2P
 RL: CAT (Catalyst use); PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(crystal structure; preparation, structural characterization, and ethylene polymerization behavior of silica supported salicylaldimine-based neutral nickel catalysts)

RN 937808-87-2 HCAPLUS

CN Nickel, [3-[[[2,6-bis(1-methylethyl)phenyl]imino- κ N]methyl][1,1'-biphenyl]-2,2'-diolato- κ O2]phenyl(triphenylphosphine)-, (SP-4-3)- (CA INDEX NAME)



CC 29-13 (Organometallic and Organometalloidal Compounds)

Section cross-reference(s): 35, 75

IT 937808-87-2P 937808-90-7P

RL: CAT (Catalyst use); PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(crystal structure; preparation, structural characterization, and ethylene polymerization behavior of silica supported salicylaldimine-based neutral nickel catalysts)

REFERENCE COUNT: 45 THERE ARE 45 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L41 ANSWER 2 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2007:345867 HCAPLUS

DOCUMENT NUMBER: 146:521917

TITLE: Substituent effects in (κ^2 -N,O)-salicylaldiminato nickel(II)-methyl pyridine polymerization catalysts: terphenyls controlling polyethylene microstructures

AUTHOR(S): Goettker-Schnetmann, Inigo; Wehrmann, Peter; Roehr, Caroline; Mecking, Stefan

CORPORATE SOURCE: Fachbereich Chemie, Universitaet Konstanz, Konstanz, D-78457, Germany

SOURCE: Organometallics (2007), 26(9), 2348-2362
CODEN: ORGND7; ISSN: 0276-7333

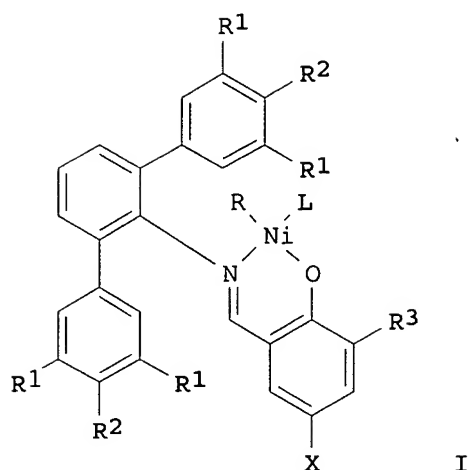
PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 146:521917

GI



AB N-Terphenyl iodo- and 9-anthracenylphenyl salicylaldimines and their methyl- and phenylnickel pyridine complexes I were prepared and evaluated for the catalytic activity in ethene polymerization; the substitution of the terphenyl group allows to control the polymer mol. weight, linearity and microstructure. A series of (κ^2 -N,O)-salicylaldiminato Ni(II)-Me pyridine complexes I (7-pyr, R3 = X = I, 8-pyr, R3 = 9-anthracenyl, X = H; R = Me, R1 = CF3, tBu, Me, MeO, R2 = H, OH, MeO) derived from 3,5-diiodosalicylaldehyde (3a) and 3-(9-anthryl)salicylaldehyde (3b), and terphenylamines 2,6-(3,5-R1-4-R2-C6H2)2C6H3-NH2 (4a, R1 = CF3, R2 = H; 4b, R1 = tBu, R2 = H; 4c, R1 = tBu, R2 = OH; 4d, R1 = Me, R2 = H; 4e, R1 = Me, R2 = MeO; 4f, R1 = MeO, R2 = H; 4g, R1 = MeO, R2 = MeO), was prepared by reaction of the resp. salicylaldimine (5a-f, 6a-g) with [(tmeda)NiMe2] (tmeda = N,N,N',N'-tetramethylethylenediamine) or [(pyridine)2NiMe2]. Complexes 7-pyr and 8-pyr are highly active single component catalysts for the polymerization of ethylene, producing a wide range of different polyethylene microstructures. While comparable complexes derived from 3a, 3b, 5-nitrosalicylaldehyde, 3-tert-butylsalicylaldehyde, 3,5-[3,5-(CF3)2C6H3]2-salicylaldehyde, and 2,6-[3,5-(CF3)2C6H3]2C6H3-NH2 afford polyethylenes with similar degrees of branching, variation of the terphenyl moieties in complexes 7-pyr and 8-pyr allows access to a wide range of polyethylene microstructures under identical reaction conditions. The x-ray diffraction analyses of complexes 7b-pyr and 8f-pyr are reported.

IT 936643-55-9P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

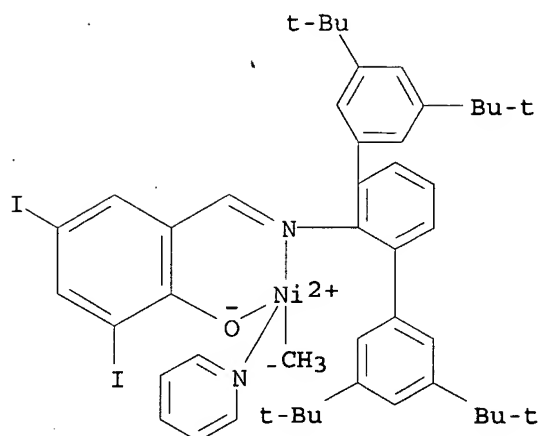
(crystal structure; preparation of nickel organometallic N-terphenyl salicylaldimine Schiff base complexes as ethene polymerization catalysts and substituent effects on polymer microstructure)

RN 936643-55-9 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(1,1-dimethylethyl)[1,1':3',1''-terphenyl]-2'-yl]imino- κ N]methyl]phenolato- κ O)methyl(pyridine)-, (SP-4-4)-, compd. with benzene (2:3) (CA INDEX NAME)

CM 1

CRN 936643-54-8
 CMF C47 H56 I2 N2 Ni O
 CCI CCS



CM 2

CRN 71-43-2
 CMF C6 H6



- CC 29-13 (Organometallic and Organometalloidal Compounds)
 Section cross-reference(s): 25, 35, 75
- IT 936643-55-9P 936643-64-0P
 RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
 (crystal structure; preparation of nickel organometallic N-terphenyl salicylaldehyde Schiff base complexes as ethene polymerization catalysts and substituent effects on polymer microstructure)
- IT 936643-54-8P 936643-63-9P
 RL: CAT (Catalyst use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)
 (mol. structure; preparation of nickel organometallic N-terphenyl salicylaldehyde Schiff base complexes as ethene polymerization catalysts and substituent effects on polymer microstructure)
- IT 680185-26-6 680185-29-9 680185-30-2
 910858-45-6
 RL: CAT (Catalyst use); USES (Uses)
 (preparation of nickel organometallic N-terphenyl salicylaldehyde Schiff base complexes as ethene polymerization catalysts and substituent effects on polymer microstructure)
- IT 936643-56-0P 936643-57-1P 936643-58-2P
 936643-59-3P 936643-60-6P 936643-61-7P
 936643-62-8P 936643-65-1P 936643-66-2P
 936643-67-3P 936643-69-5P
 RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP

(Preparation); USES (Uses)

(preparation of nickel organometallic N-terphenyl salicylaldimine Schiff base complexes as ethene polymerization catalysts and substituent effects on polymer microstructure)

REFERENCE COUNT: 48 THERE ARE 48 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L41 ANSWER 3 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2006:829497 HCAPLUS

DOCUMENT NUMBER: 145:419500

TITLE: Catalytic ethylene polymerization in carbon dioxide as a reaction medium with soluble nickel(II) catalysts

AUTHOR(S): Bastero, Amaia; Francio, Giancarlo; Leitner, Walter; Mecking, Stefan

CORPORATE SOURCE: Lehrstuhl für Chemische Materialwissenschaft, Universitaet Konstanz, Konstanz, 78457, Germany

SOURCE: Chemistry--A European Journal (2006), 12(23), 6110-6116

CODEN: CEUJED; ISSN: 0947-6539

PUBLISHER: Wiley-VCH Verlag GmbH & Co. KGaA

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 145:419500

AB A series of neutral NiII-salicylaldiminato complexes substituted with perfluorooctyl- and trifluoromethyl groups, were studied as catalyst precursors for ethylene polymerization in supercrit. CO₂. Catalyst precursors 6a and 6c, which are soluble in scCO₂, afford the highest polymer yields, corresponding to 2 x 10³ turnovers. Semicryst. polyethylene (Mn typically 104 g/mol) is obtained with variable degrees of branching (11 to 24 branches per 1000 carbon atoms, predominantly Me branches) and crystallinity (54 to 21%), depending on the substitution pattern of the catalyst.

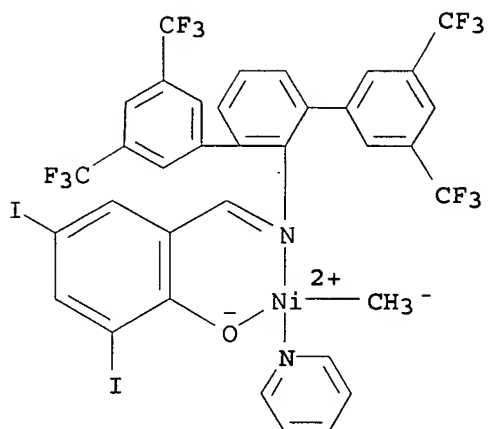
IT 680185-26-6

RL: CAT (Catalyst use); USES (Uses)

(polymerization catalyst; catalytic ethylene polymerization in carbon dioxide as reaction medium with soluble nickel catalysts)

RN 680185-26-6 HCAPLUS

CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':3',1''-terphenyl]-2'-yl]imino-κN)methyl]phenolato-κO)methyl(pyridine)-, (SP-4-4)- (9CI) (CA INDEX NAME)



CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 29, 78

IT 680185-26-6 910858-45-6

RL: CAT (Catalyst use); USES (Uses)

(polymerization catalyst; catalytic ethylene polymerization in carbon dioxide

as reaction medium with soluble nickel catalysts)

IT 912487-74-2P 912487-75-3P 912487-76-4P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP

(Preparation); USES (Uses)

(polymerization catalyst; catalytic ethylene polymerization in carbon dioxide

as reaction medium with soluble nickel catalysts)

REFERENCE COUNT: 69 THERE ARE 69 CITED REFERENCES AVAILABLE
FOR THIS RECORD. ALL CITATIONS AVAILABLE
IN THE RE FORMAT

L41 ANSWER 4 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2006:480827 HCAPLUS

DOCUMENT NUMBER: 145:158314

TITLE: Structural, spectroscopic, and electrochemical
behavior of trans-phenolato cobalt(III)
complexes of asymmetric NN'O ligands as
archetypes for metallomesogens

AUTHOR(S): Shakya, Rajendra; Imbert, Camille; Hratchian,
Hrant P.; Lanznaster, Mauricio; Heeg, Mary Jane;
McGarvey, Bruce R.; Allard, Marco; Schlegel, H.
Bernhard; Verani, Claudio N.

CORPORATE SOURCE: Department of Chemistry, Wayne State University,
Detroit, MI, 48202, USA

SOURCE: Dalton Transactions (2006), (21), 2517-2525
CODEN: DTARAF; ISSN: 1477-9226

PUBLISHER: Royal Society of Chemistry

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 145:158314

AB To understand and predict structural, redox, magnetic, and optical
properties of more complex and potentially mesogenic electroactive
comps. such as [CoIII(Lt-BuLC)2]ClO4 (1), five archetypical
complexes [CoIII(LRA)2]ClO4, where R = H (2), tert-Bu (3), methoxy
(4), nitro (5), and chloro (6), were obtained and studied by several
spectrometric, spectroscopic, and electrochem. methods. The

complexes 2, 4, and 6 were characterized by single-crystal x-ray diffraction, and show the metal center in an approx. D2h symmetry. Exptl. results support the fact that the electron donating or withdrawing nature of the phenolate-appended substituents changes dramatically the redox and spectroscopic properties of these compds. The 3d6 electronic configuration of the metal ion dominates the overall geometry adopted by these compds. with the phenolate rings occupying trans positions to one another. Formation of phenoxyl radicals was observed for 1, 3, and 6, but irreversible ligand oxidation takes place upon bulk electrolysis. These data were compared to detailed B3LYP/6-31G (d)-level computational calcns. and were used to account for the results observed. A comparison between compound 1 and archetype 3, validates the approach of using archetypical models to study metal-containing soft materials.

IT 897922-31-5P

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(preparation, cyclic voltammetry, UV-visible spectrum and electronic structure from MO calcns. as model for cobalt (pyridinylmethylaminomethyl)phenol derivative complexes)

RN 897922-31-5 HCAPLUS

CN Cobalt(1+), [2,4-bis(1,1-dimethylethyl)-6-[[(R) - [4'-(heptyloxy) [1,1'-biphenyl]-4-yl] [(2-pyridinyl-κN)methyl]amino-κN)methyl]phenolato-κO] [2,4-bis(1,1-dimethylethyl)-6-[[(S) - [4'-(heptyloxy) [1,1'-biphenyl]-4-yl] [(2-pyridinyl-κN)methyl]amino-κN)methyl]phenolato-κO]-, (OC-6-24)-, perchlorate (9CI) (CA INDEX NAME)

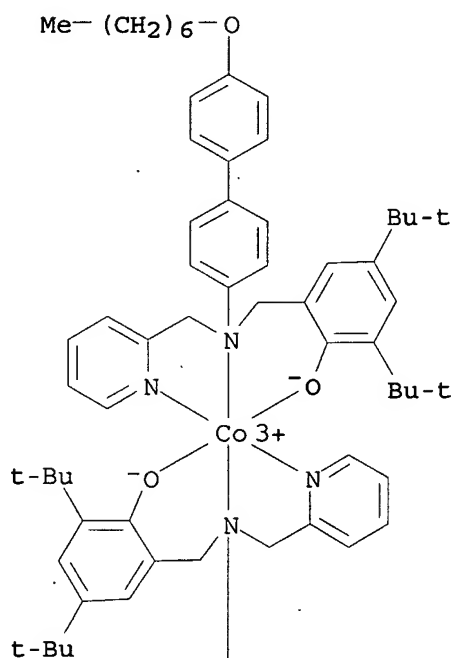
CM 1

CRN 897922-30-4

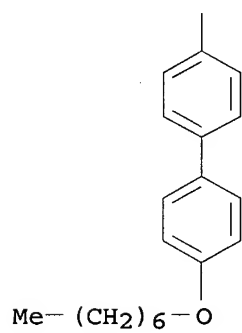
CMF C80 H102 Co N4 O4

CCI CCS

PAGE 1-A



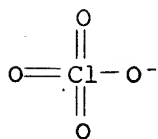
PAGE 2-A



CM 2

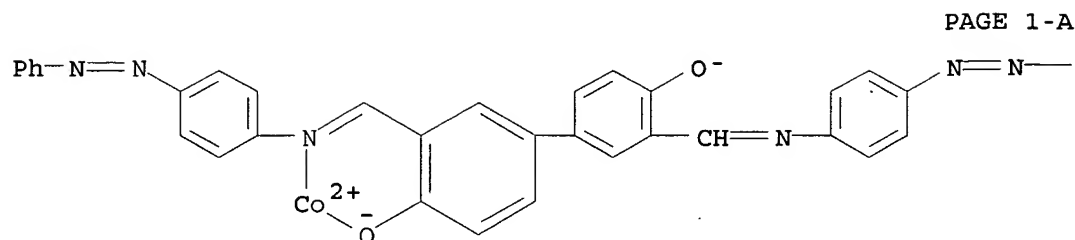
CRN 14797-73-0

CMF Cl O4



CC 78-7 (Inorganic Chemicals and Reactions)
Section cross-reference(s): 65, 72, 75
IT 897922-31-5P
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)
(preparation, cyclic voltammetry, UV-visible spectrum and electronic structure from MO calcns. as model for cobalt (pyridinylmethylaminomethyl)phenol derivative complexes)
REFERENCE COUNT: 59 THERE ARE 59 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L41 ANSWER 5 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER: 2004:396667 HCAPLUS
DOCUMENT NUMBER: 141:405121
TITLE: Synthesis and Characterization of Some Cobalt(II), Copper(II), and Nickel(II) Complexes with New Schiff Bases from the Reaction of p-Aminoazobenzene with Salicylaldehyde
AUTHOR(S): Peker, Esin; Serin, Selahattin
CORPORATE SOURCE: Faculty of Science and Arts, Department of Chemistry, Kahramanmaraş Sutcu Imam University, Kahramanmaraş, Turk.
SOURCE: Synthesis and Reactivity in Inorganic and Metal-Organic Chemistry (2004), 34(5), 859-872
CODEN: SRIMCN; ISSN: 0094-5714
PUBLISHER: Marcel Dekker, Inc.
DOCUMENT TYPE: Journal
LANGUAGE: English
OTHER SOURCE(S): CASREACT 141:405121
AB Two new Schiff bases were synthesized by the reaction of p-aminoazobenzene with salicylaldehyde ((p-salicylideneamino)azobenzene, L1H) and the oxidative polycondensation product of L1H in which the 5-positions of the salicylaldehyde rings are directly bonded (L2H2). The complexes of cobalt(II), copper(II), and nickel(II) with L1H and L2H2 were prepared. The ligands and their complexes were characterized by elemental analyses, IR spectra, electronic absorption spectra, mass spectra, ¹H and ¹³C NMR spectra, and magnetic susceptibility measurements. The thermal properties of all complexes were studied by TG and DTA. The anal. data show that the metal to ligand ratio in the L1H complexes is 1:2, but in the oligomeric L2H2 complexes it is 1:1.
IT 787624-17-3P
RL: SPN (Synthetic preparation); PREP (Preparation)
(oligomeric; preparation of cobalt(II), copper(II) and nickel(II) mononuclear and oligomeric complexes of Schiff bases from aminoazobenzene and salicylaldehyde)
RN 787624-17-3 HCAPLUS
CN Cobalt, [3-[[[4-(phenylazo)phenyl]imino-κN)methyl]-3'-[[[4-(phenylazo)phenyl]imino]methyl][1,1'-biphenyl]-4,4'-diolato(2-)-κO4]- (9CI) (CA INDEX NAME)

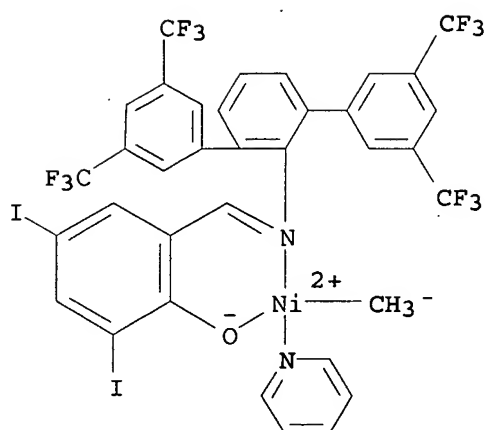


PAGE 1-B

— Ph

CC 78-7 (Inorganic Chemicals and Reactions)
 IT 787624-17-3P 787624-18-4P 787624-19-5P
 RL: SPN (Synthetic preparation); PREP (Preparation)
 (oligomeric; preparation of cobalt(II), copper(II) and nickel(II)
 mononuclear and oligomeric complexes of Schiff bases from
 aminoazobenzene and salicylaldehyde)
 REFERENCE COUNT: 22 THERE ARE 22 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L41 ANSWER 6 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2004:158319 HCAPLUS
 DOCUMENT NUMBER: 140:339707
 TITLE: Remote substituents controlling catalytic
 polymerization by very active and robust neutral
 nickel(II) complexes
 AUTHOR(S): Zuideveld, Martin A.; Wehrmann, Peter; Roehr,
 Caroline; Mecking, Stefan
 CORPORATE SOURCE: Institut fuer Makromolekulare Chemie und
 Freiburger Material-forschungszentrum,
 Albert-Ludwig-Universitaet Freiburg, Freiburg,
 79104, Germany
 SOURCE: Angewandte Chemie, International Edition (2004),
 43(7), 869-873
 CODEN: ACIEF5; ISSN: 1433-7851
 PUBLISHER: Wiley-VCH Verlag GmbH & Co. KGaA
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 AB Remote control: Substituents strongly affect the catalytic
 properties of complexes 1 in ethylene polymerization, despite their
 remoteness from the active center. An appropriate substitution
 pattern provides active and robust catalysts.
 IT 680185-26-6P
 RL: CAT (Catalyst use); PRP (Properties); SPN (Synthetic
 preparation); PREP (Preparation); USES (Uses)
 (crystal structure; preps. of robust neutral nickel(II)
 complexes for catalytic ethylene polymerization)
 RN 680185-26-6 HCAPLUS
 CN Nickel, [2,4-diiodo-6-[[[3,3'',5,5''-tetrakis(trifluoromethyl)[1,1':
 3',1''-terphenyl]-2'-yl]imino-κN)methyl]phenolato-
 κO)methyl(pyridine)-, (SP-4-4)- (9CI) (CA INDEX NAME)



CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 67, 75, 78

IT 680185-26-6P 680185-28-8P

RL: CAT (Catalyst use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)
(crystal structure; preps. of robust neutral nickel(II) complexes for catalytic ethylene polymerization)

IT 332951-99-2P 680185-27-7P 680185-29-9P

680185-30-2P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(preps. of robust neutral nickel(II) complexes for catalytic ethylene polymerization)

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L41 ANSWER 7 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2002:459784 HCAPLUS

DOCUMENT NUMBER: 137:194513

TITLE: Synthesis, thermal, electrical and biological studies of some Schiff base complexes

AUTHOR(S): Bhawe, N. S.; Bahad, P. J.; Sonparote, P. M.; Aswar, A. S.

CORPORATE SOURCE: Department of Chemistry, Nagpur University, Nagpur, 440 010, India

SOURCE: Journal of the Indian Chemical Society (2002), 79(4), 342-344

CODEN: JICSAH; ISSN: 0019-4522

PUBLISHER: Indian Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 137:194513

AB Complexes of CrIII, MnIII, FeIII and AlIII with Schiff bases derived from the 2:1 condensation of 2-hydroxy-5-methylacetophenone and 2-hydroxy-5-chloroacetophenone with p-phenylenediamine (H2L1/H2L2) were prepared. The antibacterial activities of the ligands and their complexes were evaluated. The kinetic parameters of decomposition also were evaluated by both Freeman-Carroll and Sharp-Wentworth methods. D.C. elec. conductivity of the complexes was studied.

IT 451491-21-7P

RL: BSU (Biological study, unclassified); PRP (Properties); RCT

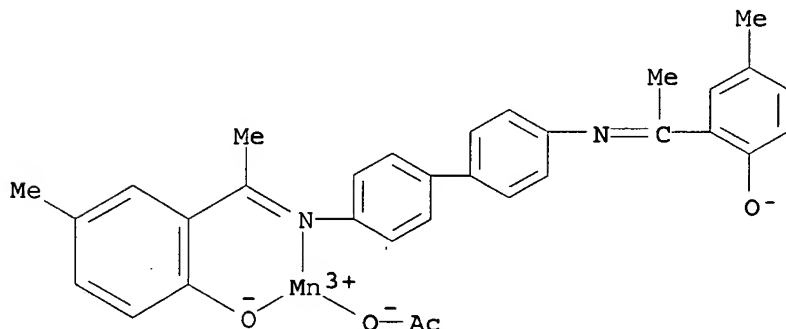
(Reactant); SPN (Synthetic preparation); BIOL (Biological study);
PREP (Preparation); RACT (Reactant or reagent)

(preparation, elec. conductivity, thermal decomposition kinetics and
antibacterial

activity of polymeric)

RN 451491-21-7 HCAPLUS

CN Manganese, (acetato- κ O) [2-[1-[[4'-[[1-(2-hydroxy-5-methylphenyl)ethylidene]amino][1,1'-biphenyl]-4-yl]imino- κ N]ethyl]-4-methylphenolato(2-)- κ O]- (CA INDEX NAME)



CC 78-7 (Inorganic Chemicals and Reactions)

Section cross-reference(s): 10, 67, 76

IT 451491-18-2P 451491-20-6P 451491-21-7P

451491-23-9P 451491-24-0P 451491-26-2P

451491-29-5P 451491-32-0P

RL: BSU (Biological study, unclassified); PRP (Properties); RCT
(Reactant); SPN (Synthetic preparation); BIOL (Biological study);

PREP (Preparation); RACT (Reactant or reagent)

(preparation, elec. conductivity, thermal decomposition kinetics and
antibacterial

activity of polymeric)

REFERENCE COUNT: 15 THERE ARE 15 CITED REFERENCES AVAILABLE
FOR THIS RECORD. ALL CITATIONS AVAILABLE
IN THE RE FORMAT

L41 ANSWER 8 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1994:644089 HCAPLUS

DOCUMENT NUMBER: 121:244089

TITLE: Mononuclear and binuclear iron(III) complexes of
(-N2O2-) tetradentate Schiff base ligands

AUTHOR(S): Gaber, M.; Issa, R. M.; Ghoniem, M. M.;
El-Baradie, K. Y.

CORPORATE SOURCE: Fac. Sci., Tanta Univ., Tanta, Egypt

SOURCE: Egyptian Journal of Chemistry (1992), Volume
Date 1991, 34(2), 107-19

CODEN: EGJCA3; ISSN: 0367-0422

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The Fe³⁺ chelates of some N2O2-Schiff bases, derived from
salicylaldehyde, 2,4-dihydroxybenzaldehyde or 2-hydroxy-1-
naphthaldehyde and aromatic diamines were studied by conductance
measurements to gain information about their probable stoichiometry.
The solid complexes were isolated and studied by TGA, IR, electronic
and ESR spectra, magnetic moment determination and elemental anal.
Mononuclear and binuclear complexes are formed. The Fe³⁺ showed

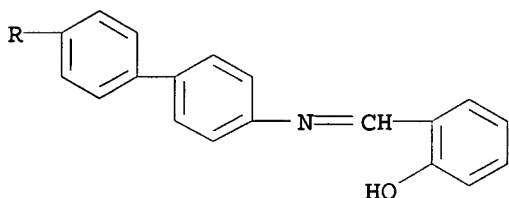
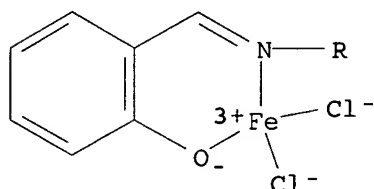
octahedral configuration. The values of the magnetic moments showed a possible antiferromagnetic effect.

IT 158334-13-5P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation and IR spectra and thermal decomposition of)

RN 158334-13-5 HCAPLUS

CN Iron, [[2,2'-[[1,1'-biphenyl]-4,4'-diylbis(nitrilomethylidyne)]bis(p
henolato)](1-)-N,O]dichloro- (9CI) (CA INDEX NAME)



CC 78-7 (Inorganic Chemicals and Reactions)

IT 122474-54-8P 122474-56-0P 137931-08-9P 137953-82-3P
158334-04-4P 158334-08-8P 158334-09-9P 158334-10-2P
158334-11-3P 158334-12-4P 158334-13-5P 158334-14-6P
158334-15-7P 158334-16-8P 158334-17-9P 158334-18-0P
158334-19-1P 158334-20-4P 158334-21-5P 158367-05-6P
158367-06-7P 158367-07-8P 158367-08-9P 158367-09-0P
158367-10-3P 158367-11-4P 158367-12-5P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation and IR spectra and thermal decomposition of)

L41 ANSWER 9 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1994:423338 HCAPLUS

DOCUMENT NUMBER: 121:23338

TITLE: Synthesis and mesomorphic properties of some
platinum(II) and oxovanadium(IV) complexes

AUTHOR(S): Sadashiva, B. K.; Ghode, Archana

CORPORATE SOURCE: Raman Res. Inst., Bangalore, 560080, India

SOURCE: Liquid Crystals (1994), 16(1), 33-42

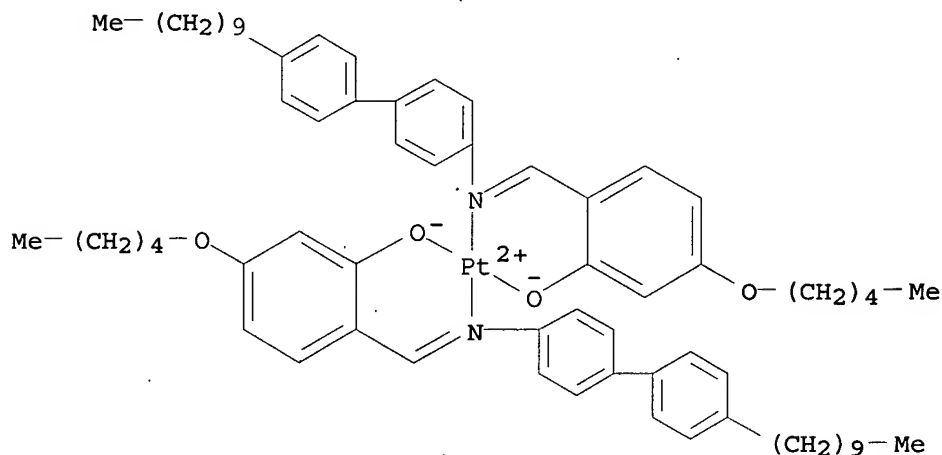
CODEN: LICRE6; ISSN: 0267-8292

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The synthesis and mesomorphic properties of a homologous series of N-(2-hydroxy-4-n-alkoxybenzylidene)-4''-n-decylphenylanilines and their Pt(II) and oxovanadium(IV) complexes are reported. All the ligands and their metal chelates exhibit enantiotropic mesophases, predominantly smectic A and smectic C phases. The transition temps. and enthalpies were determined for most of the compds. The Pt(II) complexes have higher m.ps. and mesophase thermal stabilities. However, the oxovanadium(IV) complexes have a wider thermal range for the mesophase. Both Pt(II) and oxovanadium(IV) complexes containing only a chain on the biphenyl moiety exhibit a nematic phase.

IT 155476-41-8P
 RL: PRP (Properties); SPN (Synthetic preparation); PREP
 (Preparation)
 (preparation and liquid crystal properties of)
 RN 155476-41-8 HCAPLUS
 CN Platinum, bis[2-[[[4'-decyl[1,1'-biphenyl]-4-yl]imino]methyl]-5-(pentyloxy)phenolato-N2,O1]-, (SP-4-1)- (9CI) (CA INDEX NAME)



CC 75-11 (Crystallography and Liquid Crystals)
 Section cross-reference(s): 25, 78
 IT 155097-91-9P 155097-92-0P 155097-93-1P 155097-94-2P
 155097-95-3P 155097-96-4P 155097-97-5P 155097-98-6P
 155097-99-7P 155098-00-3P 155098-01-4P 155098-02-5P
 155098-03-6P 155476-32-7P 155476-33-8P 155476-34-9P
 155476-35-0P 155476-36-1P 155476-37-2P 155476-38-3P
 155476-39-4P 155476-40-7P 155476-41-8P
 155476-42-9P 155476-43-0P 155476-44-1P
 155476-45-2P 155476-46-3P 155476-47-4P
 155476-48-5P 155476-49-6P 155515-40-5P
 155677-74-0P
 RL: PRP (Properties); SPN (Synthetic preparation); PREP
 (Preparation)
 (preparation and liquid crystal properties of)

L41 ANSWER 10 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1994:335697 HCAPLUS

DOCUMENT NUMBER: 120:335697

TITLE: Mesogenic properties of nickel(II) complexes of N-(2-hydroxy-4-n-alkyloxybenzylidene)-4''-n-dodecylphenylanilines

AUTHOR(S): Prasad, Veena; Sadashiva, B. K.

CORPORATE SOURCE: Raman Res. Inst., Bangalore, 560080, India

SOURCE: Molecular Crystals and Liquid Crystals Science and Technology, Section A: Molecular Crystals and Liquid Crystals (1994), 241, 167-74
 CODEN: MCLCE9; ISSN: 1058-725X

DOCUMENT TYPE: Journal

LANGUAGE: English

AB Ni(II) complexes of N-(2-hydroxy-4-n-alkyloxybenzylidene)-4''-n-dodecylphenylanilines were prepared. These predominantly exhibit smectic A and smectic C phases. A comparison of the mesomorphic

behavior of these complexes with those of the corresponding Cu(II) and Pd(II) complexes was made. The Pd complexes have the highest thermal stability of the mesophases while the Ni complexes have the lowest m.ps. These Ni complexes are diamagnetic in the solid state and exhibit paramagnetic properties in CHCl₃ solution

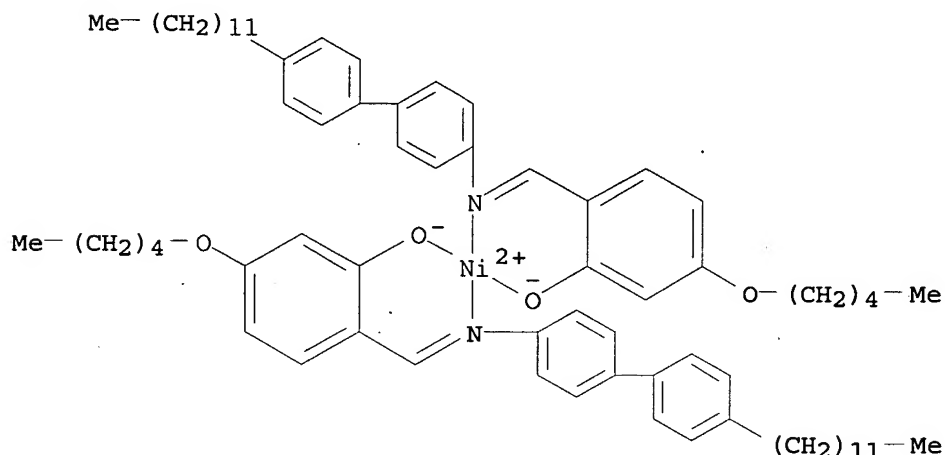
IT 155515-78-9P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(liquid crystal, preparation and properties of)

RN 155515-78-9 HCAPLUS

CN Nickel, bis[2-[[[4'-dodecyl[1,1'-biphenyl]-4-yl]imino]methyl]-5-(pentyloxy)phenolato-N2,O1]- (9CI) (CA INDEX NAME)



CC 75-11 (Crystallography and Liquid Crystals)

Section cross-reference(s): 78

IT 155515-78-9P 155515-79-0P 155515-80-3P

155515-81-4P 155515-82-5P 155515-83-6P

155515-84-7P 155515-85-8P 155515-86-9P

155515-87-0P 155515-88-1P 155515-89-2P

155515-90-5P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(liquid crystal, preparation and properties of)

L41 ANSWER 11 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1994:91379 HCAPLUS

DOCUMENT NUMBER: 120:91379

TITLE: Liquid crystalline properties of o-hydroxy substituted Schiff's bases and their copper(II) and palladium(II) complexes

AUTHOR(S): Prasad, Veena; Sadashiva, B. K.

CORPORATE SOURCE: Raman Res. Inst., Bangalore, 560080, India

SOURCE: Molecular Crystals and Liquid Crystals Science and Technology, Section A: Molecular Crystals and Liquid Crystals (1993), 225, 303-12
CODEN: MCLCE9; ISSN: 1058-725X

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The synthesis and liquid crystalline properties of thirty nine compds. are reported. These include a homologous series of N-(2-hydroxy-4-n-alkoxybenzylidene)-4''-n-dodecylphenylanilines and their

corresponding Cu(II) and Pd(II) complexes. DSC and optical microscopy were used to characterize the mesophases. The Schiff's bases exhibit polymesomorphism while their metal chelates basically show smectic A and smectic C phases except for a few derivs. The transition temps. of the metal chelates are considerably higher than the ligands from which they are derived.

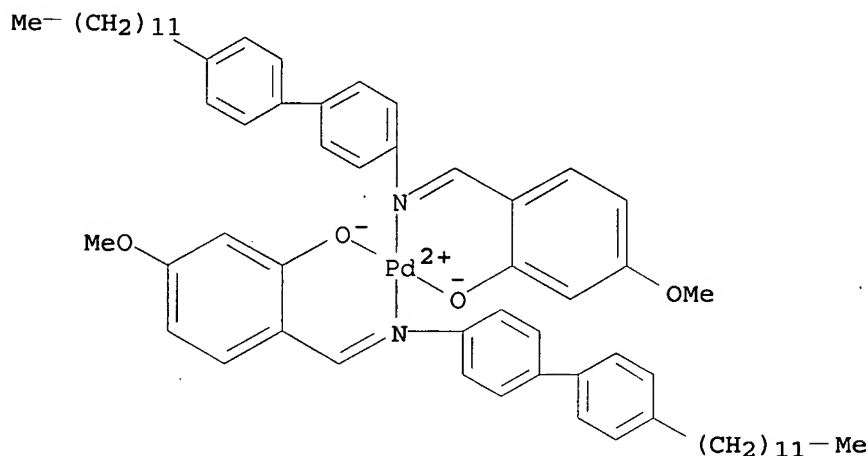
IT 152690-65-8P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(liquid crystal, preparation and transition temps. of)

RN 152690-65-8 HCAPLUS

CN Palladium, bis[2-[[[(4'-dodecyl[1,1'-biphenyl]-4-yl)imino]methyl]-5-methoxyphenolato-N2,O1]- (9CI) (CA INDEX NAME)



CC 75-11 (Crystallography and Liquid Crystals)

Section cross-reference(s): 78

IT 152501-84-3P 152501-85-4P 152501-86-5P 152501-87-6P
 152501-88-7P 152501-89-8P 152501-90-1P 152501-91-2P
 152501-92-3P 152501-93-4P 152501-94-5P 152501-95-6P
 152501-96-7P 152690-40-9P 152690-41-0P 152690-42-1P
 152690-43-2P 152690-44-3P 152690-45-4P 152690-46-5P
 152690-47-6P 152690-48-7P 152690-61-4P 152690-62-5P
 152690-63-6P 152690-64-7P 152690-65-8P
 152690-66-9P 152690-67-0P 152690-68-1P
 152690-69-2P 152690-70-5P 152690-71-6P
 152690-72-7P 152690-73-8P 152690-74-9P
 152690-75-0P 152690-76-1P 152690-77-2P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(liquid crystal, preparation and transition temps. of)

L41 ANSWER 12 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1990:244765 HCAPLUS

DOCUMENT NUMBER: 112:244765

TITLE: Polymeric coordination complexes of some bipoisitive metal ions with Schiff bases

AUTHOR(S): Dwivedi, D. K.; Shukla, R. K.; Maurya, R. C.

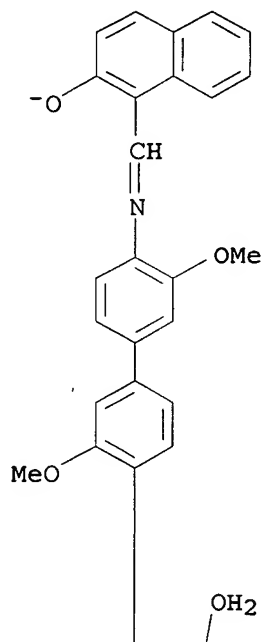
CORPORATE SOURCE: Chem. Lab., Atarra Coll., Atarra, India

SOURCE: Acta Ciencia Indica, Chemistry (1988), 14(2), 91-8

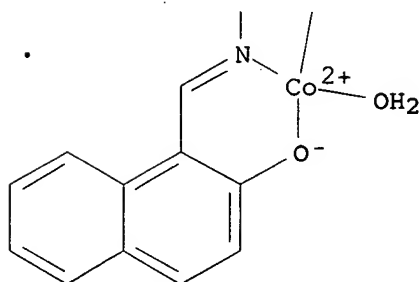
CODEN: ACICDV; ISSN: 0253-7338

DOCUMENT TYPE: Journal
 LANGUAGE: English
 AB [ML₂H₂O]_n (M = Mn, Co, Ni, Cu, Zn; H₂L = bis(2-hydroxy-1-naphthylmethylene)-o-dianisidine, -p-phenylenediamine, -benzidine) and H₂L were prepared. The Schiff bases are tetradentate bridging in the polymeric compds. The crystal field parameters were calculated for the Co, Ni and Cu complexes.
 IT 122436-27-5P
 RL: SPN (Synthetic preparation); PREP (Preparation) (preparation and electronic and IR spectra and crystal field parameters of)
 RN 122436-27-5 HCAPLUS
 CN Cobalt, diaqua[[1,1'-[(3,3'-dimethoxy[1,1'-biphenyl]-4,4'-diyl)bis(nitrilomethylidyne)]bis[2-naphthalenolato]](2-)-N1,O2]-(9CI) (CA INDEX NAME)

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CC 78-7 (Inorganic Chemicals and Reactions)
IT 122436-27-5P 122436-28-6P 122436-29-7P
122436-31-1P 122436-32-2P 122436-35-5P
122469-35-6P 127261-28-3P 127261-29-4P
RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation and electronic and IR spectra and crystal field
parameters of)
IT 122436-30-0P 122436-34-4P 122517-08-2P
RL: PRP (Properties); SPN (Synthetic preparation); PREP
(Preparation)
(preparation and electronic and IR spectra of)

L41 ANSWER 13 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1990:209744 HCAPLUS

DOCUMENT NUMBER: 112:209744

TITLE: Interaction of manganese(II) Schiff's base
complexes with sulfur dioxide molecules

AUTHOR(S): Abu-El Wafa, Samy M.; Issa, Raafat M.

CORPORATE SOURCE: Fac. Educ., Ain Shams Univ., Roxy, Egypt

SOURCE: Bulletin de la Societe Chimique de France
(1989), (Sept.-Oct.), 595-8

CODEN: BSCFAS; ISSN: 0037-8968

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The preparation of [Mn(SB)SO₂] [H₂SB = tetradentate o-HOC₆H₄CH:N)2Z (Z =
(CH₂)₂, (CH₂)₃, (CH₂)₄, o-C₆H₄, p-C₆H₄, p-C₆H₄C₆H₄), (RCH:N)2Z (R =
2-hydroxynaphthyl)] is reported. The complexes are characterized by
elemental anal., IR, electronic spectra and conductance
measurements. TGA of the complexes indicated that the SO₂ mol. is
thermally stable <150-160°C after which it is evolved and the
complex decomps. The effective magnetic moments at room temperature of
the complexes is 5.71-5.40 μB/Mn atom. ESR spectra of the
complexes in CHCl₃ solution at 300 and 93 K suggest octahedral geometry
around the Mn(II) ion.

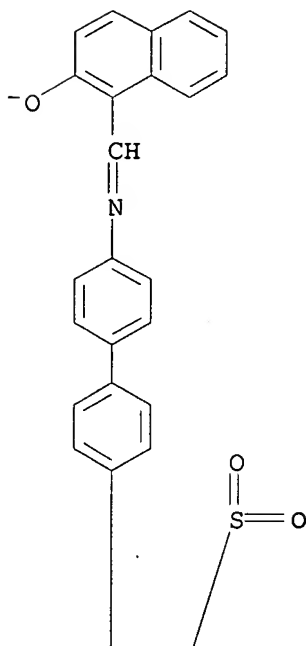
IT 126824-70-2P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation and ESR and IR spectrum of)

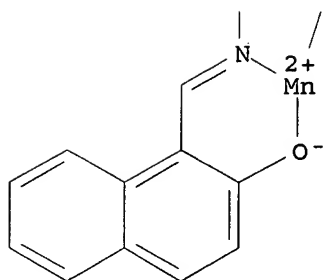
RN 126824-70-2 HCAPLUS

CN Manganese, [[1,1'-[[1,1'-biphenyl]-4,4'-
diylbis(nitrilomethylidyne)]bis[2-naphthalenolato]](2-)-N,O] (sulfur
dioxide-S)- (9CI) (CA INDEX NAME)

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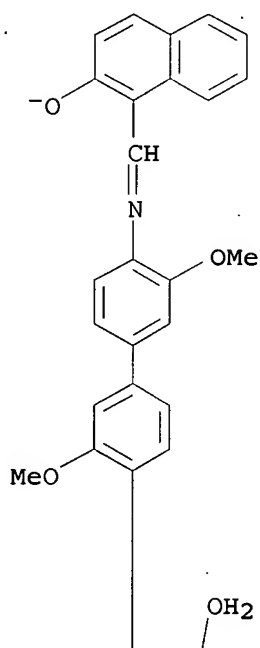


CC 78-7 (Inorganic Chemicals and Reactions)
 IT 126824-70-2P
 RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation and ESR and IR spectrum of)
 IT 126773-22-6P 126796-63-2P 126796-64-3P 126796-65-4P
 126796-66-5P 126796-67-6P 126796-69-8P 126796-73-4P
 126824-69-9P
 RL: PRP (Properties); SPN (Synthetic preparation); PREP
 (Preparation)
 (preparation and IR spectrum of)

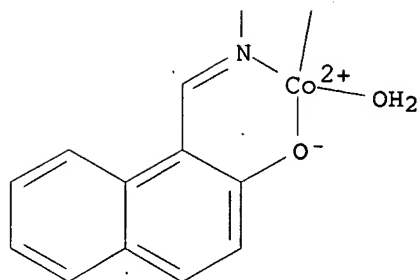
L41 ANSWER 14 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1989:566094 HCAPLUS
 DOCUMENT NUMBER: 111:166094
 TITLE: Coordination-polymers of manganese(II),
 cobalt(II), nickel(II), copper(II), and zinc(II)

with some novel Schiff bases
AUTHOR(S): Dwivedi, D. K.
CORPORATE SOURCE: Chem. Lab., Atarra Coll., Atarra, 210 201, India
SOURCE: Current Science (1988), 57(22), 1238-41
CODEN: CUSCAM; ISSN: 0011-3891
DOCUMENT TYPE: Journal
LANGUAGE: English
AB Polymeric $ML(H_2O)_2$ ($M = Co, Cu, Mn, Ni, Zn$; H_2L = Schiff bases from 2-hydroxy-1-naphthaldehyde and aryldiamines) were prepared and characterized by UV spectra and ligand field parameter determination and magnetic moments. Metal coordination is octahedral.
IT 122436-27-5P
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
(preparation and crystal field parameters of)
RN 122436-27-5 HCAPLUS
CN Cobalt, diaqua[[1,1'-[(3,3'-dimethoxy[1,1'-biphenyl]-4,4'-diyl)bis(nitrilomethylidyne)]bis[2-naphthalenolato]](2-)-N1,O2]-(9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 2-A



CC 78-7 (Inorganic Chemicals and Reactions)
 IT 122436-27-5P 122436-28-6P 122436-29-7P
 122436-31-1P 122436-32-2P 122436-33-3P 122436-35-5P
 122436-36-6P 122469-35-6P

RL: PRP (Properties); SPN (Synthetic preparation); PREP
 (Preparation)

(preparation and crystal field parameters of)

IT 3172-37-0P 16196-98-8P 17635-31-3P 122436-30-0P
 122436-34-4P 122517-08-2P 123143-58-8P

RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation of)

L41 ANSWER 15 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1989:164996 HCAPLUS

DOCUMENT NUMBER: 110:164996

TITLE: Structural, semiconducting and thermal studies
 of some Schiff-base coordination polymers

AUTHOR(S): Aswar, A. S.; Bahad, P. J.; Pardhi, A. V.;
 Bhawe, N. S.

CORPORATE SOURCE: Dep. Chem., Nagpur Univ., Nagpur, 440 010, Niger
 SOURCE: Journal of Polymer Materials (1988), 5(4), 233-9

CODEN: JOPME8; ISSN: 0970-0838

DOCUMENT TYPE: Journal

LANGUAGE: English

AB Schiff base coordination polymers of Co(II), Ni(II), Cu(II), and
 Zn(II) with substituted acetophenone and aromatic diamine such as
 bis(2-hydroxy-5-methylacetophenone)benzidine (I) and
 bis(2-hydroxy-5-chloroacetophenone)benzidine (II) were prepared I and
 II were characterized by elemental anal., IR, and NMR. Strong
 interamol. H bonding (O-H...N) of phenolic OH with amine N was observed
 as indicated by IR and NMR spectra. The coordination polymers were
 characterized by elemental anal., magnetic, spectral (electronic and
 IR) and thermal and elec. conductivity studies. On the basis of these
 anal., a structure for these polymers was suggested.

IT 120015-55-6P

RL: PRP (Properties); SPN (Synthetic preparation); PREP
 (Preparation)

(preparation and elec. conductivity of)

RN 120015-55-6 HCAPLUS

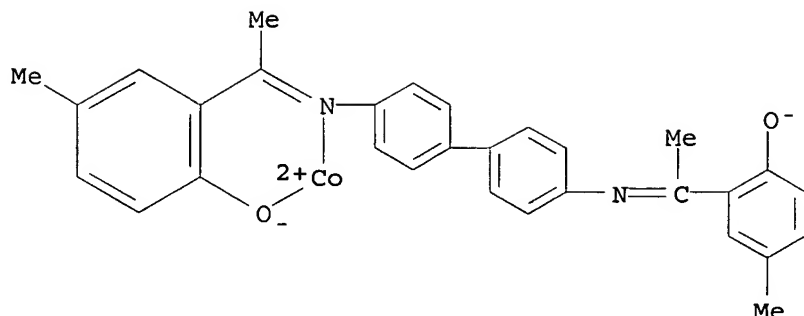
CN Cobalt, [[2,2'-[[[1,1'-biphenyl]-4,4'-diylbis(nitriloethylidyne)]bis[
 4-methylphenolato]](2-)-N,O]-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 120015-54-5

CMF C30 H26 Co N2 O2

CCI CCS



CC 78-7 (Inorganic Chemicals and Reactions)
 Section cross-reference(s): 25, 76
 IT 120015-55-6P 120015-59-0P 120015-63-6P
 120028-94-6P
 RL: PRP (Properties); SPN (Synthetic preparation); PREP
 (Preparation)
 (preparation and elec. conductivity of)
 IT 119986-72-0P 119986-73-1P 120015-57-8P
 120015-61-4P 120015-65-8P 120015-67-0P
 RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation of)

L41 ANSWER 16 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1988:618698 HCAPLUS

DOCUMENT NUMBER: 109:218698

TITLE: X-ray K absorption spectroscopic study of some Schiff-base polymers of cobalt

AUTHOR(S): Mujumdar, P. V.; Deshmukh, Prabodhachandra C.; Sapre, V. B.; Bhawe, N. S.

CORPORATE SOURCE: Dep. Phys., Nagpur Univ., Nagpur, India

SOURCE: X-Ray Spectrometry (1988), 17(4), 133-7

CODEN: XRSPAX; ISSN: 0049-8246

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The results of a study of the x-ray K absorption spectrum of Co in the elemental form and in some of its Schiff-base coordination polymers are presented. The K absorption discontinuity of Co was recorded together with the XANES and EXAFS associated with it by employing a 0.4 m Cauchois-type bent-crystal spectrograph. The (100) planes of a muscovite mica analyzer crystal were used for reflection of x-radiation, the source of which was the W target of a conventional x-ray tube. Measurements were made of the chemical shift, edge width, XANES, and EXAFS associated with the discontinuity. R. M. Levy's theory (1965) and the graphical method based on the theory of E. A. Stern et al. were employed to determine the average radius of the 1st coordination sphere surrounding the Co atom in all the systems studied. The results are discussed in terms of the bond length, bond ionicity, coordination geometry, and some other parameters related to the local environment of the absorbing Co atom.

IT 117564-62-2

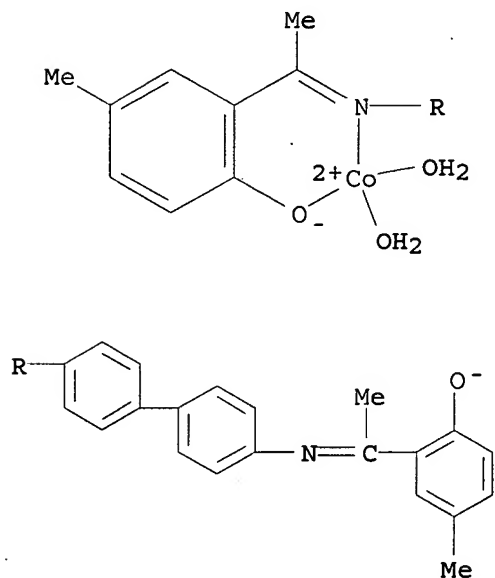
RL: PRP (Properties)

(x-ray K absorption spectra of cobalt in)

RN 117564-62-2 HCAPLUS

CN Cobalt, diaqua[[2,2'-[[1,1'-biphenyl]-4,4'-

diylbis(nitriloethylidyne)]bis[4-methylphenolato]](2-)-N2,O1] - (9CI)
(CA INDEX NAME)



CC 73-6 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

IT 117564-62-2 117584-70-0 117608-80-7

RL: PRP (Properties)

(x-ray K absorption spectra of cobalt in)

L41 ANSWER 17 OF 28 HCAPLUS, COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1987:648920 HCAPLUS

DOCUMENT NUMBER: 107:248920

TITLE: Polymeric coordination complexes of some bipovalent metal ions with Schiff bases derived from 5-bromosalicylaldehyde and p-phenylenediamine or benzidine

AUTHOR(S): Dwivedi, D. K.; Maurya, R. C.; Shukla, R. K.; Anandam, N.; Shukla, R.; Dubey, R. C.

CORPORATE SOURCE: Chem. Lab., Atarra Coll., Atarra, 210 201, India

SOURCE: Oriental Journal of Chemistry (1987), 3(2), 160-4

CODEN: OJCHEG; ISSN: 0970-020X

DOCUMENT TYPE: Journal

LANGUAGE: English

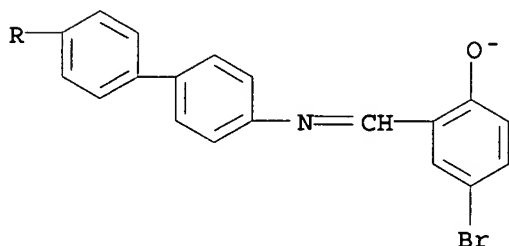
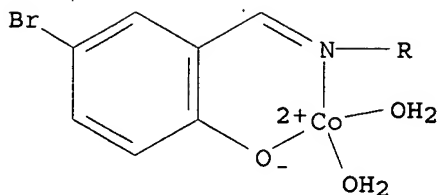
AB [ML(H₂O)₂]_n (M = Co, Ni, Cu; H₂L = bis(5-bromosalicylidene)-p-phenylenediamine and -benzidine) were prepared and characterized on the basis of anal., magnetic, thermogravimetric and spectral data. The Schiff bases behave as dibasic tetradentate ONNO donor ligands.

IT 111570-36-6P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation and IR spectrum and crystal field parameters of polymeric)

RN 111570-36-6 HCAPLUS

CN Cobalt, diaqua[[2,2'-[[1,1'-biphenyl]-4,4'-diylbis(nitrilomethylidyne)]bis[4-bromophenolato]](2-)-N,O] - (9CI)
(CA INDEX NAME)



CC 78-7 (Inorganic Chemicals and Reactions)
 IT 111549-10-1P 111570-34-4P 111570-35-5P 111570-36-6P
 111570-37-7P 111619-10-4P
 RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation and IR spectrum and crystal field parameters of
 polymeric)

L41 ANSWER 18 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1985:596493 HCAPLUS
 DOCUMENT NUMBER: 103:196493
 TITLE: Synthesis and stereochemical studies of some new
 chelate polymers
 AUTHOR(S): Patel, M. N.; Jani, B. N.
 CORPORATE SOURCE: Dep. Chem., Sardar Patel Univ., Vallabh
 Vidyanagar, 388120, India
 SOURCE: Journal of Macromolecular Science, Chemistry
 (1985), A22(11), 1517-26
 CODEN: JMCHBD; ISSN: 0022-233X
 DOCUMENT TYPE: Journal
 LANGUAGE: English

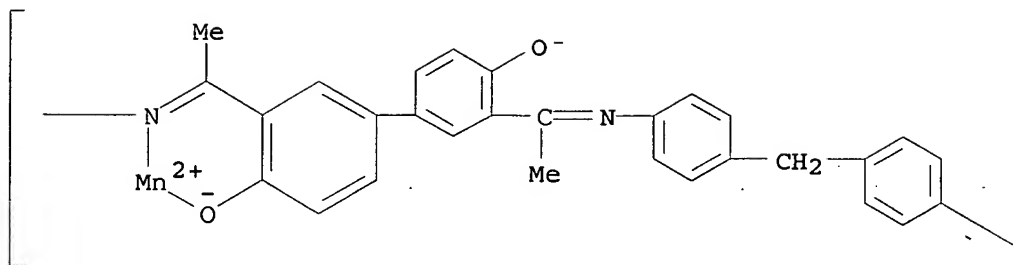
AB The polychelates of divalent metal ions such as Cu, Ni, Co, Mn, and
 Zn were synthesized from polymeric ligands derived from
 4,4'-dihydroxy-3,3'-diacetylbiphenyl and 4,4'-
 diaminodiphenylmethane. Elemental anal., magnetic susceptibility,
 visible and IR spectra, and thermal properties of the polychelates
 were studied. The anal. data suggested the general formula (ML)_n (M
 = metal, L = ligand) for the polychelates. The polychelates were
 quite stable but insol. in common organic solvents.

IT 98767-31-8P
 RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation, spectra and magnetic properties of)

RN 98767-31-8 HCAPLUS

CN Poly[nitriloethylidyne(4,4'-dihydroxy[1,1'-biphenyl]-3,3'-
 diyl)ethylidynenitrilo-1,4-phenylenemethylene-1,4-phenylene
 manganese complex] (9CI) (CA INDEX NAME)

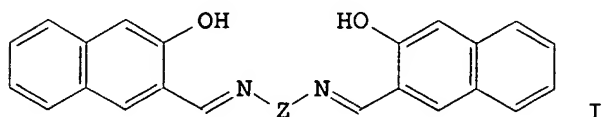
PAGE 1-A



PAGE 1-B

CC 35-5 (Chemistry of Synthetic High Polymers)
 Section cross-reference(s): 36
 IT 7439-96-5DP, complexes with hydroxy-containing polymeric Schiff bases
 7440-02-0DP, complexes with hydroxy-containing polymeric Schiff bases
 7440-48-4DP, complexes with hydroxy-containing polymeric Schiff bases
 7440-50-8DP, complexes with hydroxy-containing polymeric Schiff bases
 7440-66-6DP, complexes with hydroxy-containing polymeric Schiff bases
 98573-05-8DP, complexes with divalent metals 98767-30-7P
98767-31-8P 98767-32-9P 98767-33-0P
 RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation, spectra and magnetic properties of)

L41 ANSWER 19 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1985:105062 HCAPLUS
 DOCUMENT NUMBER: 102:105062
 TITLE: New manganese(II) and copper(II) complexes of
 tetradentate N2O2 Schiff base ligands
 AUTHOR(S): Abu-El-Wafa, S. M.; Ashmawy, F. M.; Issa, R. M.;
 McAuliffe, C. A.; Parish, R. V.
 CORPORATE SOURCE: Fac. Educ., Ain Shams Univ., Cairo, Egypt
 SOURCE: Inorganica Chimica Acta (1985), 96(1), L25-L27
 CODEN: ICHAA3; ISSN: 0020-1693
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 GI



AB $MnL(H_2O)_2$ ($H_2L = I$, $Z = (CH_2)_2$, $o-C_6H_4C_6H_4-o$; $H_2L = o-HOC_6H_4CH:NZN:CHC_6H_4OH-o$, $Z = o-C_6H_4C_6H_4-o$) and $CuL(H_2O)$ were prepared and characterized by elemental anal., magnetic moment, molar conductivity, ESR, and IR and visible reflectance spectral methods. The H_2O mols. are coordinated and L_2- is tetradentate via imino N and hydroxy O. A cis monomeric structure is suggested for $MnL(H_2O)_2$ ($H_2L = I$, $Z = (CH_2)_2$) and a polymeric structure for the other complexes.

IT 94901-33-4P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

RN 94901-33-4 HCAPLUS

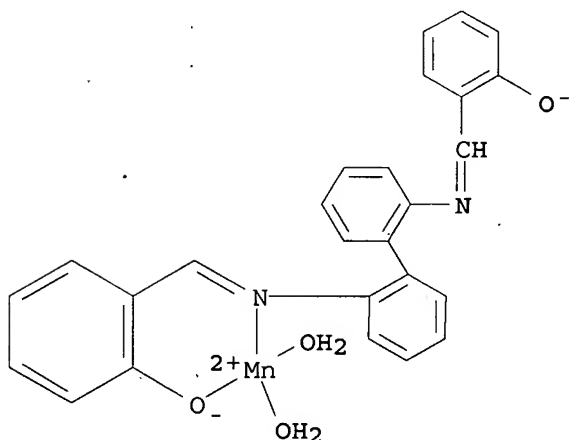
CN Manganese, diaqua[[2,2'-[[1,1'-biphenyl]-2,2'-diylbis(nitrilomethylidyne)]bis[phenolato]](2-)-N,O]-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 94901-32-3

CMF C26 H22 Mn N2 O4

CCI CCS



CC 78-7 (Inorganic Chemicals and Reactions)

IT 94901-31-2P 94901-33-4P 94955-48-3P 94955-50-7P

94976-55-3P 94976-56-4P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

L41 ANSWER 20 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1982:192331 HCAPLUS

DOCUMENT NUMBER: 96:192331

ORIGINAL REFERENCE NO.: 96:31537a,31540a

TITLE: Novel chain polymers by condensation of metal(II) coordinated salicylaldehyde with diamines

AUTHOR(S): Maurya, Pyare L.; Agarwala, Badri V.; Dey, Arun K.

CORPORATE SOURCE: Chem. Lab., Univ. Allahabad, Allahabad, 211 002, India

SOURCE: Makromolekulare Chemie (1982), 183(3), 511-16
CODEN: MACEAK; ISSN: 0025-116X

DOCUMENT TYPE: Journal

LANGUAGE: English

AB Coordination polymers, $-(ML)_n-$ were synthesized by condensation of bis(salicylaldehyde)metal(II) ($M = Co, Ni, \text{ and } Cu$) with diamines (1,4-diaminobenzene and 4,4'-diaminobiphenyl). Their composition, structure, thermal stability, and phys. properties were investigated by anal., magnetic, spectral (IR and diffuse reflectance spectra), and thermogravimetric studies. In the bis(salicylaldehyde)metal(II)-chelates, 2 salicylaldehyde mols. are linked to each metal ion through the O atoms of the phenolic and aldehydic groups and in the polymeric complexes the N atoms of the amino groups replace the aldehyde oxygens. The diamine mol. is thus acting as a bridging unit between 2 salicylaldehydrometal(II) units. The polymers obtained from 4,4'-diaminophenyl are thermally more stable than those from 1,4-diaminobenzene, the order of thermal stability in both cases being $Ni(II) > Cu(II) > Co(II)$. $Co(II)$ and $Cu(II)$ complexes are paramagnetic, whereas $Ni(II)$ complexes are diamagnetic. The reflectance spectra along with the magnetic data suggest a square planar structure for $Cu(II)$ and $Ni(II)$ complexes, and a tetrahedral one for the $Co(II)$ complexes.

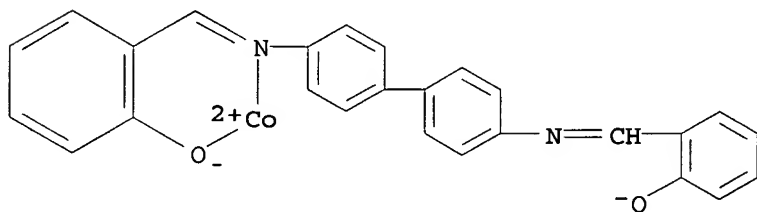
IT 52658-78-3P
RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

RN 52658-78-3 HCAPLUS

CN Cobalt, [[2,2'-[[1,1'-biphenyl]-4,4'-diylbis(nitrilomethylidyne)]bis[phenolato]](2-)]-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 52658-77-2
CMF C26 H18 Co N2 O2
CCI CCS



CC 78-7 (Inorganic Chemicals and Reactions)

IT 29932-09-0P 52658-78-3P 52682-48-1P 59926-65-7P
59926-69-1P 81613-05-0P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

L41 ANSWER 21 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1981:454006 HCAPLUS

DOCUMENT NUMBER: 95:54006

ORIGINAL REFERENCE NO.: 95:9007a,9010a

TITLE: Cobalt(II) complexes of quadridentate poly Schiff bases

AUTHOR(S): Patel, M. N.; Patil, Shirish H.

CORPORATE SOURCE: Dep. Chem., Sardar Patel Univ., Vallabh

SOURCE: Vidyanagar, 388 120, India
 Indian Journal of Chemistry, Section A:
 Inorganic, Physical, Theoretical & Analytical
 (1981), 20A(5), 523-4
 CODEN: IJCADU; ISSN: 0376-4710

DOCUMENT TYPE: Journal

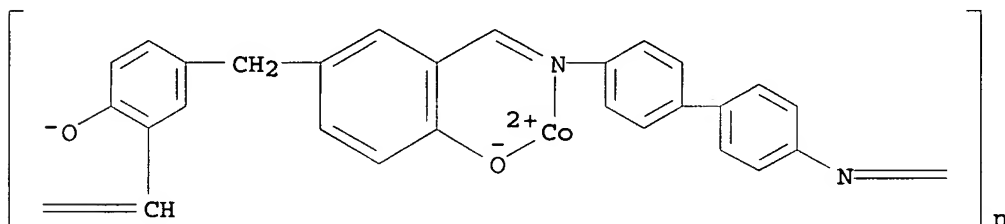
LANGUAGE: English

AB Co(II) complexes of Schiff bases polymers derived from
 5,5'-methylenebis(salicylaldehyde) or 5,5'-
 sulfonylbis(salicylaldehyde) and various diamines are reported.
 Tetrahedral structures are assigned to the chelates on the basis of
 anal., electronic spectral, and magnetic susceptibility data.

IT 78332-32-8P
 RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation of)

RN 78332-32-8 HCAPLUS

CN Poly[nitrilo[1,1'-biphenyl]-4,4'-diyl-nitrilomethylidyne (6-hydroxy-
 1,3-phenylene)methylene (4-hydroxy-1,3-phenylene)methylidyne cobalt
 complex] (9CI) (CA INDEX NAME)



CC 78-7 (Inorganic Chemicals and Reactions)

IT 78332-30-6P 78332-31-7P 78332-32-8P 78332-33-9P
 78332-34-0P 78350-72-8P

RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation of)

L41 ANSWER 22 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1976:455951 HCAPLUS

DOCUMENT NUMBER: 85:55951

ORIGINAL REFERENCE NO.: 85:8981a,8984a

TITLE: Complexes of some bivalent metals with
 tetradentate Schiff bases

AUTHOR(S): Chikina, N. L.; Litvinov, V. V.; Kolodyazhnyi,
 Yu. V.; Kolomin, L. G.; Osipov, O. A.

CORPORATE SOURCE: Rostov. Gos. Univ., Rostov, USSR

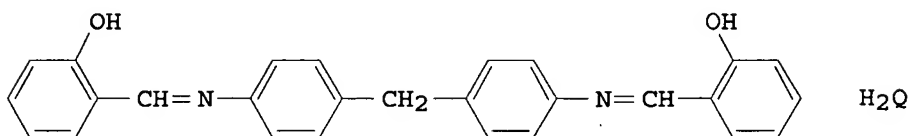
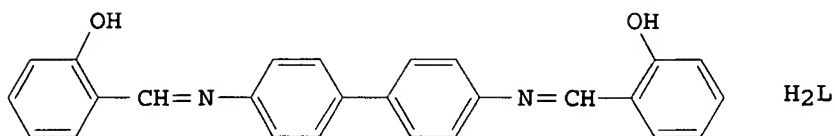
SOURCE: Koordinatsionnaya Khimiya (1976), 2(5), 632-7

CODEN: KOKHDC; ISSN: 0132-344X

DOCUMENT TYPE: Journal

LANGUAGE: Russian

GI



AB M(OAc)₂ (M = Zn, Cu, Co, Ni, Be) and the Schiff bases, H₂L and H₂Q, react in refluxing DMF to give ML and MQ, resp. The insoly. and the magnetic moments of the ML complexes indicate that the complexes are coordination polymers and the central metal has a tetrahedral environment. The decrease in the elec. conductivity and the increase in the activation energy in the transition from Cu to Ni to Co to Zn indicate that the basic contribution to the conductivity of the samples is due to the electronic states and not to ionic structure. The Zn and Be complexes luminesce under the effect of uv radiation in the solid and liquid phases.

IT 52658-78-3P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
(preparation and properties of)

RN 52658-78-3 HCAPLUS

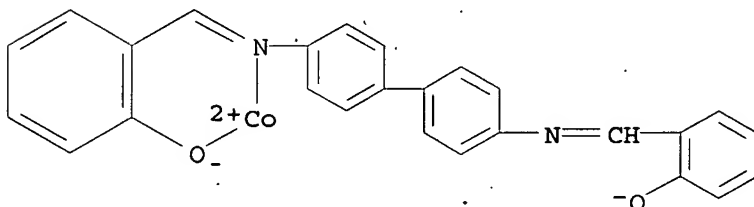
CN Cobalt, [[2,2'-[[1,1'-biphenyl]-4,4'-diylbis(nitrilomethylidene)]bis[phenolato]](2-)]-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 52658-77-2

CMF. C26 H18 Co N2 O2

CCI CCS



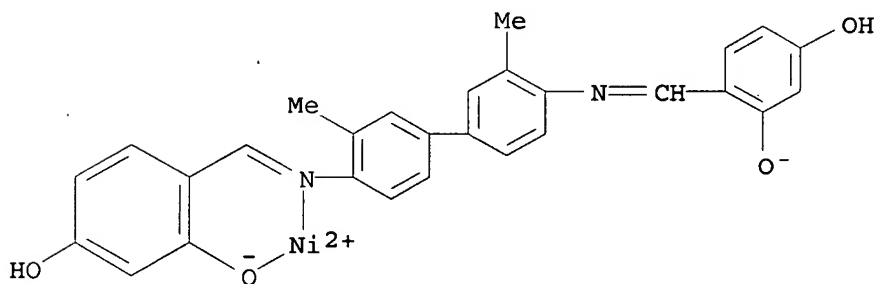
CC 78-7 (Inorganic Chemicals and Reactions)

IT 36344-60-2P 52658-78-3P 59926-65-7P 59926-67-9P
59926-69-1P 59926-71-5P 59926-73-7P 59926-75-9P
59926-77-1P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
(preparation and properties of)

L41 ANSWER 23 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER: 1975:132470 HCAPLUS

DOCUMENT NUMBER: 82:132470
 ORIGINAL REFERENCE NO.: 82:21113a,21116a
 TITLE: The ir spectra of transition metal polymeric complexes with new Schiff bases
 AUTHOR(S): Doadrio, A.; Craciunescu, D.; Shohet, Y.
 CORPORATE SOURCE: Fac. Pharm., Univ. Madrid, Madrid, Spain
 SOURCE: Farmaseuttinen Aikakauslehti (1974), 83(3), 157-61
 CODEN: FMAKAZ; ISSN: 0367-259X
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 AB ML (M = Cu²⁺, Co²⁺, Ni²⁺, Zn; and H₂L = Schiff bases derived from 2,4-dihydroxybenzaldehyde and 3,3'-dimethyl-4,4'-diaminodiphenylmethane, 3,3'-dimethyl-4,4'-diaminodicyclohexylmethane, 4,4'-diaminodicyclohexylmethane, 3,3'-dimethylbenzidine) were prepared and characterized by is spectra. The complexes are dimeric or polymeric. The Schiff bases act as tetradentate ligands.
 IT 55543-46-9P
 RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)
 RN 55543-46-9 HCAPLUS
 CN Nickel, [[4,4'-[(3,3'-dimethyl[1,1'-biphenyl]-4,4'-diyl)bis(nitrilomethylidene)]bis[1,3-benzenediolato]](2-)-N₄O₃]-, homopolymer (9CI) (CA INDEX NAME)
 CM 1
 CRN 55543-45-8
 CMP C28 H22 N2 Ni O4
 CCI CCS

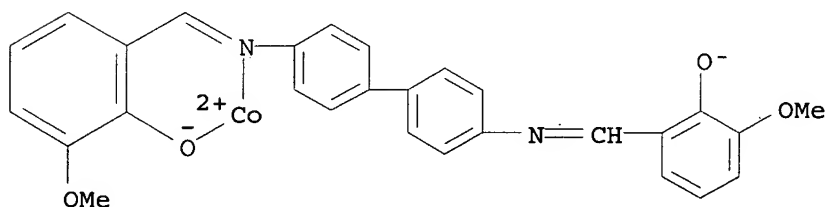


CC 78-7 (Inorganic Chemicals and Reactions)
 IT 55543-44-7P 55543-46-9P 55543-48-1P 55543-50-5P
 55543-52-7P 55571-64-7P 55571-66-9P 55571-68-1P 55571-70-5P
 55571-72-7P 55607-75-5P
 RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

L41 ANSWER 24 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1974:88890 HCAPLUS
 DOCUMENT NUMBER: 80:88890
 ORIGINAL REFERENCE NO.: 80:14287a,14290a
 TITLE: High-spin cobalt(II) chelates with tetradentate Schiff bases
 AUTHOR(S): Zelentsov, V. V.; Bogdanov, A. P.; Rukhadze, E. G.; Talyzenkova, G. P.

CORPORATE SOURCE: Mosk. Fiz.-Tekh. Inst., Moscow, USSR
 SOURCE: Zhurnal Neorganicheskoi Khimii (1973), 18(12),
 3350-3
 CODEN: ZNOKAQ; ISSN: 0044-457X
 DOCUMENT TYPE: Journal
 LANGUAGE: Russian
 AB To determine the structure of Co(II) salicylalazine and vanillalazine
 complexes, the magnetic susceptibility and the magnetic moment were
 determined at 80-300°K. Taking into account the value of a
 magnetic moment that corresponds to the $S = 3/2$ state of Co(II),
 data on the magnetic moment of these compds. indicate that the
 salicylalazine complexes have a chain-like (polymeric) structure
 composed of N2O2 tetrahedrons, whereas the structure of the
 vanillalazine complexes is probably octahedral.
 IT 52658-39-6
 RL: PRP (Properties)
 (magnetic properties of)
 RN 52658-39-6 HCAPLUS
 CN Cobalt, [[2,2'-[[1,1'-biphenyl]-4,4'-diylbis(nitrilomethylidyne)]bis
 [6-methoxyphenolato]](2-)-N2,O1]-, homopolymer. (9CI) (CA INDEX
 NAME)
 CM 1
 CRN 52658-38-5
 CMF C28 H22 Co N2 O4
 CCI CCS



CC 72-1 (Magnetic Phenomena)
 IT 52658-37-4 52658-39-6 52658-74-9 52658-76-1
 52658-78-3 52678-88-3 52682-48-1
 RL: PRP (Properties)
 (magnetic properties of)

L41 ANSWER 25 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1973:521378 HCAPLUS
 DOCUMENT NUMBER: 79:121378
 ORIGINAL REFERENCE NO.: 79:19677a,19680a
 TITLE: Coordination polymers of Schiff base ligands and
 their monomeric analogs
 AUTHOR(S): Dunski, Naphthali; Crawford, Thomas H.
 CORPORATE SOURCE: Dep. Chem., Univ. Louisville, Louisville, KY,
 USA
 SOURCE: Journal of Inorganic and Nuclear Chemistry
 (1973), 35(8), 2707-17
 CODEN: JINCAO; ISSN: 0022-1902
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 AB Polymeric (ML)_x [M = Ni, Co, Cu; H₂L = N,N'-bis(salicylidene)-3,3'-
 bis(methylthio)benzidine, N,N'-[bis(salicylidene)]-bis(o-

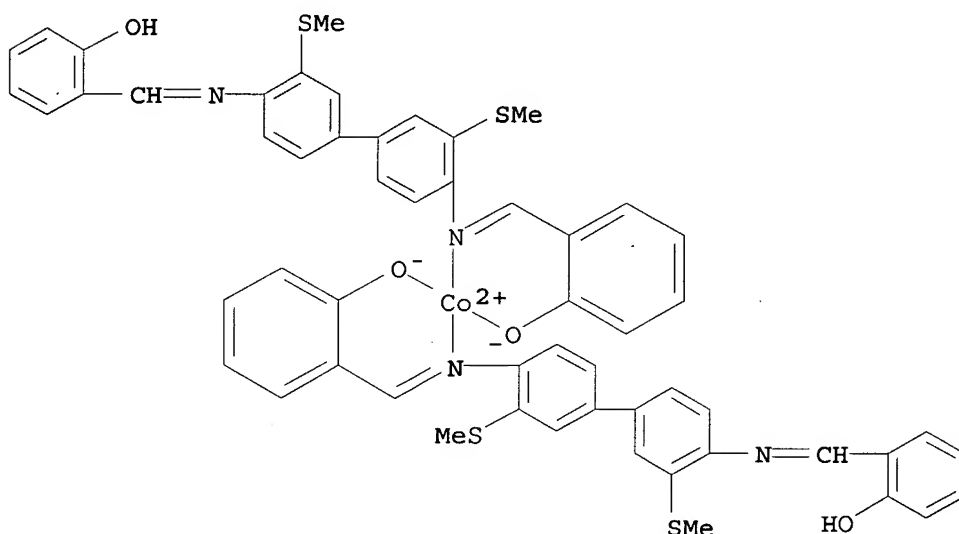
methylthioaniline)] and their monomeric analogs ML12 (H2L1 = N-salicylidene-o-methylthioaniline, M = Ni, Co, Cu; H2L1 = N-(o-hydroxyacetophenylidene)-o-methylthioaniline, M = Ni, Cu) were prepared and characterized by anal., spectra, and magnetic properties. Pd2LCl2 and PdLCl were prepared, no polymeric Pd(II) complexes were isolated. Monomeric CoL2 [HL = N,N'-bis(salicylidene)-3,3'-bis(methylthio)benzidine] was low-spin, with approx. square planar coordination.

IT 42992-90-5P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

RN 42992-90-5 HCAPLUS

CN Cobalt, bis[[2,2'-[[3,3'-bis(methylthio)[1,1'-biphenyl]-4,4'-diyl]bis(nitrilomethylidyne)]bis[phenolato]](1-)]-, (SP-4-1)- (9CI)
(CA INDEX NAME)



CC 78-7 (Inorganic Chemicals and Reactions)

IT 19850-36-3P 42805-84-5P 42805-85-6P 42805-86-7P 42871-23-8P
42871-24-9P 42871-25-0P 42871-26-1P 42871-27-2P 42942-58-5P
42942-59-6P 42947-43-3P 42992-89-2P 42992-90-5P
42993-30-6P 42993-31-7P 42993-32-8P 42993-33-9P 42993-34-0P
42993-35-1P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

L41 ANSWER 26 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1972:93950 HCAPLUS

DOCUMENT NUMBER: 76:93950

ORIGINAL REFERENCE NO.: 76:15065a,15068a

TITLE: Diphenyl derivative complexes. XXVI. Complexes of bivalent metals with the Schiff bases of aromatic diamines and β -resorcyraldehyde
AUTHOR(S): Macarovici, Const. Gh.; Dorutiu, A.; Gal, M.
CORPORATE SOURCE: Univ. Babes-Bolyai, Cluj, Rom.
SOURCE: Studia Universitatis Babes-Bolyai, Chemia (1971), 16(2), 31-7
CODEN: SUBCAB; ISSN: 1224-7154

DOCUMENT TYPE:

Journal

LANGUAGE: French

AB New complex chelates were prepared by reaction of acetates of Co, Ni, Cu, and Zn with Schiff bases of bis(4-aminophenyl)methane, 2,7-diaminofluorene, and 4,4'-diaminobiphenyl with β -resorcyaldehyde. The metal/Schiff base ratio is 1:1 for all the complexes. From the IR spectra it is concluded that the complexes, which do not melt at $\leq 300^\circ$, are probably dimers or polymers.

IT 36251-07-7P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

RN 36251-07-7 HCAPLUS

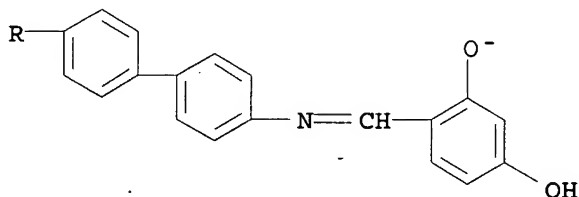
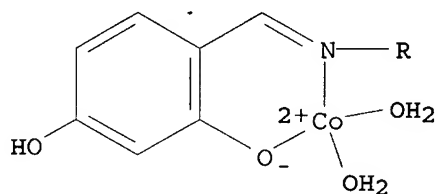
CN Cobalt, diaqua[[4,4'-[[1,1'-biphenyl]-4,4'-diylbis(nitrilomethylidyne)]bis[1,3-benzenediolato]](2-)-N4,O3]-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 47765-98-0

CMF C26 H22 Co N2 O6

CCI CCS



CC 78 (Inorganic Chemicals and Reactions)

IT 36250-99-4P 36251-00-0P 36251-01-1P 36251-02-2P 36251-03-3P

36251-04-4P 36251-05-5P 36251-06-6P 36251-07-7P

36251-08-8P 36352-09-7P 36352-10-0P 36352-11-1P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

L41 ANSWER 27 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1959:75781 HCAPLUS

DOCUMENT NUMBER: 53:75781

ORIGINAL REFERENCE NO.: 53:13706i,13707a-c

TITLE: Magnetic measurements on some complexes of
nickel with organic bases

AUTHOR(S): Manca, Pietrino

CORPORATE SOURCE: Univ. Cagliari, Italy

SOURCE: Rend. seminar. fac. sci. univ. Cagliari (1958),
28, 69-75

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

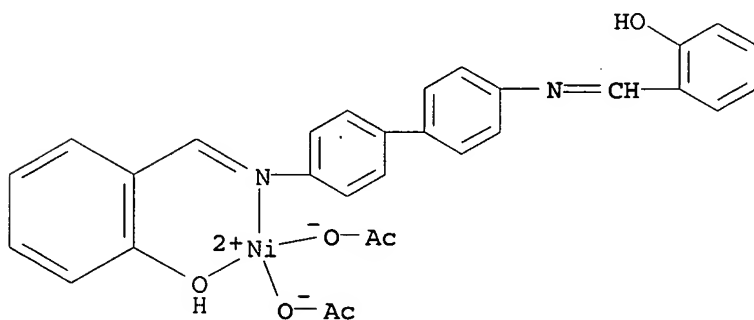
AB In the course of related work Ni complexes with amines and Schiff bases were prepared: Ni bromosalicylaldehyde (I), Ni bromosalicylaldehyde-o-phenylenediamine (II), Ni salicylaldehyde-benzidine acetate (III), and Ni ethylenediamine sulfate (IV). Methods of preparation and results of elementary analysis are given for each. Magnetic measurements were made at 291°K. by use of the Weiss-Foex-Forrer type translation pendulum. II is diamagnetic. For I, III, and IV, resp., $\chi_{sp} + 10^{-5} = 0.818, 0.796, 0.143$; $\chi_{\mu} + 10^{-3} = 4.119, 4.860, 4.900$; Bohr magnetons 3.11, 3.36, 3.38 vs. 2.83 theoretical; each have 2 unpaired electrons. The exptl. data were corrected for the diamagnetism of the nonmetallic portion of the mols. Diamagnetism of II is presumed due to dsp² hybridization of Ni electrons, giving rise to Ni²⁺ ion in the complex with a d⁸sp² configuration. In I the tetrahedral sp³ Ni hybrid gives rise to Ni²⁺ in d⁸sp³ configuration and a tetrahedral complex, in apparent agreement with other Ni analogs (Tyson and Adams, C.A. 34, 43148). The structure of III is attributed tentatively to the formation of a dimer complex with Ni tetrahedrally coordinated by 2 CH₃COO⁻ and 2 N. III is completely insol. in most of the common organic solvents. The Ni of IV is octahedrally coordinated by 3 ethylenediamines in 4 ionic, partly covalent, structures in resonance, each of which yields a net of 2 unpaired electrons accounting for the paramagnetism. IV is considered similar to Ni²⁺ coordinated by 6 H₂O but with greater energy of ligands.

IT 129153-11-3

(Derived from data in the 6th Collective Formula Index (1957-1961))

RN 129153-11-3 HCAPLUS

CN Nickel, bis(acetato-κO) [2-[[[4'-[[[2-hydroxyphenyl)methylene]amino][1,1'-biphenyl]-4-yl]imino-κN]methyl]phenol-κO]- (CA INDEX NAME)



CC 2 (General and Physical Chemistry)

IT 15304-51-5 125737-61-3 127185-95-9 129153-11-3

(Derived from data in the 6th Collective Formula Index (1957-1961))

L41 ANSWER 28 OF 28 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1959:75780 HCAPLUS

DOCUMENT NUMBER: 53:75780

ORIGINAL REFERENCE NO.: 53:13706h-i

TITLE: Magnetic properties at low temperatures of bivalent fluorides of cobalt and nickel

AUTHOR(S): Bizette, H.; Tsai, B.

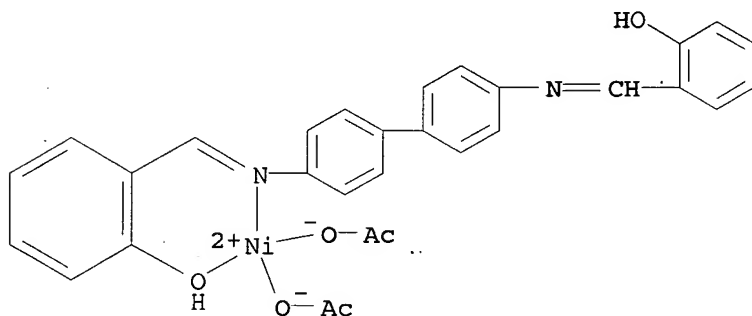
SOURCE: Bull. inst. intern. froid Annexe (1955) 149-52.
 DOCUMENT TYPE: Journal
 LANGUAGE: Unavailable

AB New results on the magnetic susceptibility in the temperature range from 4 to 300°K. are reported for CoF₂ and NiF₂. CoF₂ follows the Curie-Weiss law with the expression $\chi_M = 3.00/(T + 37)$. An inflection occurs at 37.7°K., where there is a maximum in the heat capacity. NiF₂ shows metamagnetism at low temps. and below 73°K., the location of a maximum in the heat capacity, the magnetization depends on whether or not the sample is cooled in a magnetic field. The results for both removed the apparent contradiction between magnetic properties and neutron-diffraction results.

IT 129153-11-3
 (Derived from data in the 6th Collective Formula Index (1957-1961))

RN 129153-11-3 HCAPLUS

CN Nickel, bis(acetato-κO) [2-[[[4'-[[[2-hydroxyphenyl)methylene]amino][1,1'-biphenyl]-4-yl]imino-κN]methyl]phenol-κO]- (CA INDEX NAME)



CC 2 (General and Physical Chemistry)

IT 15304-51-5 125737-61-3 127185-95-9 129153-11-3
 (Derived from data in the 6th Collective Formula Index (1957-1961))

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